

THEO13 (Oct. 28, 2013, CERN)

Thorium Energy R&D in China

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Jialuo Road 2019, Jiading, Shanghai 201800, China

Outline

Why TMSR

Th-E @ TMSR

Status and Progress



Outline

Why TMSR

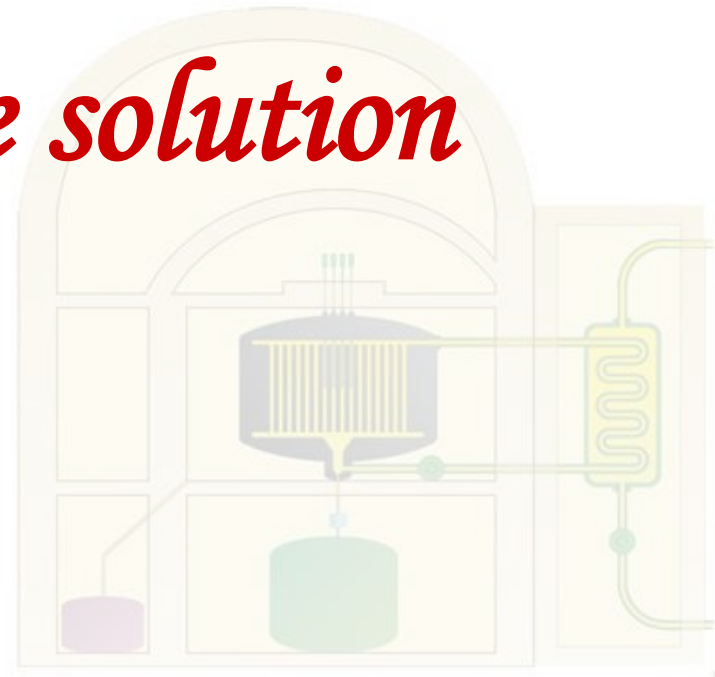
Th-E @ TMSR

Status and Progress



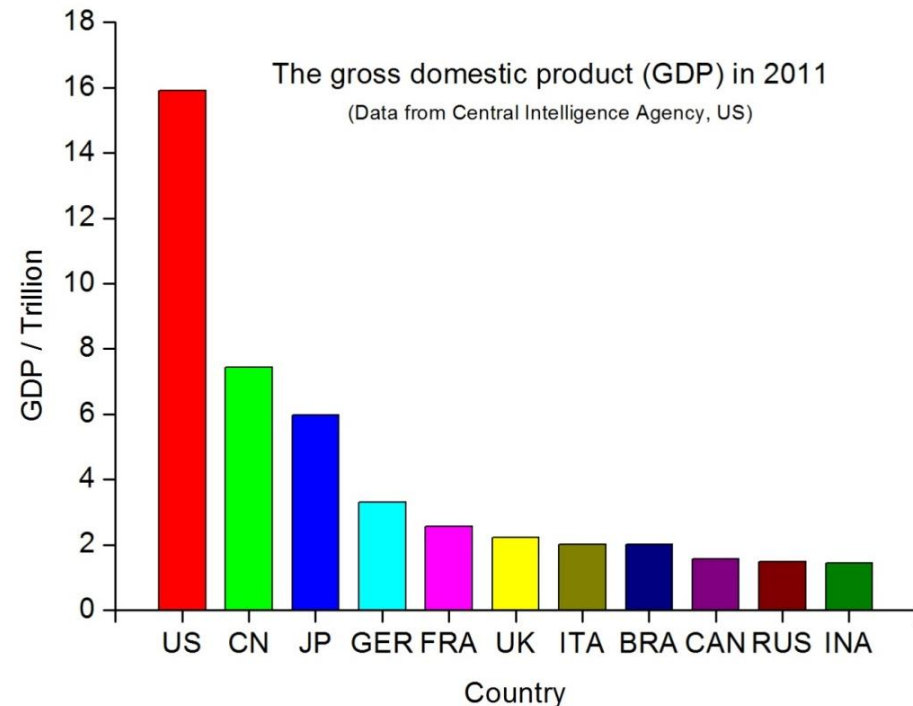
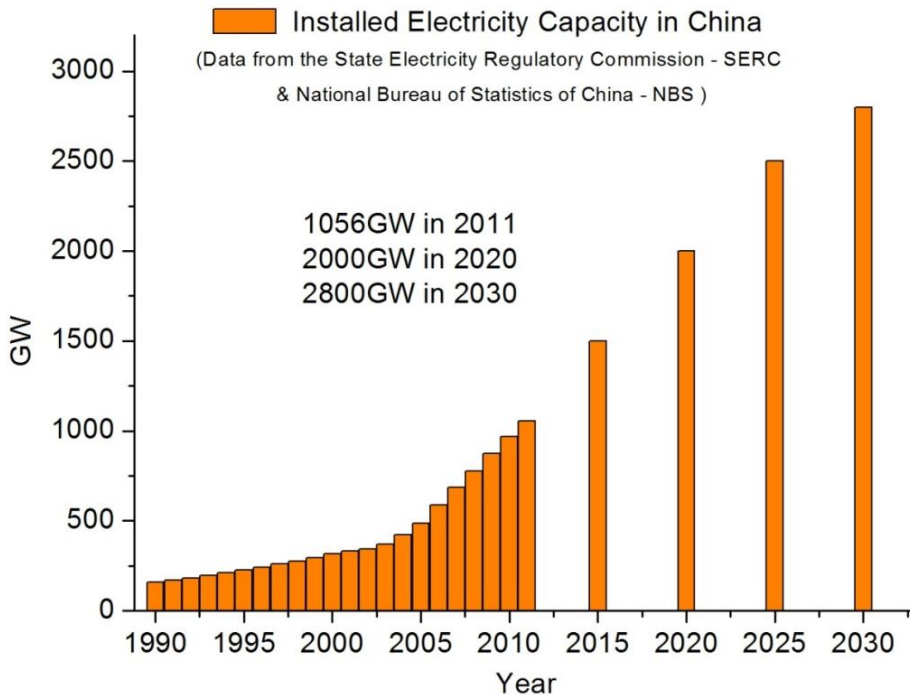
The Challenges to China

And one possible solution



China's Energy Challenge

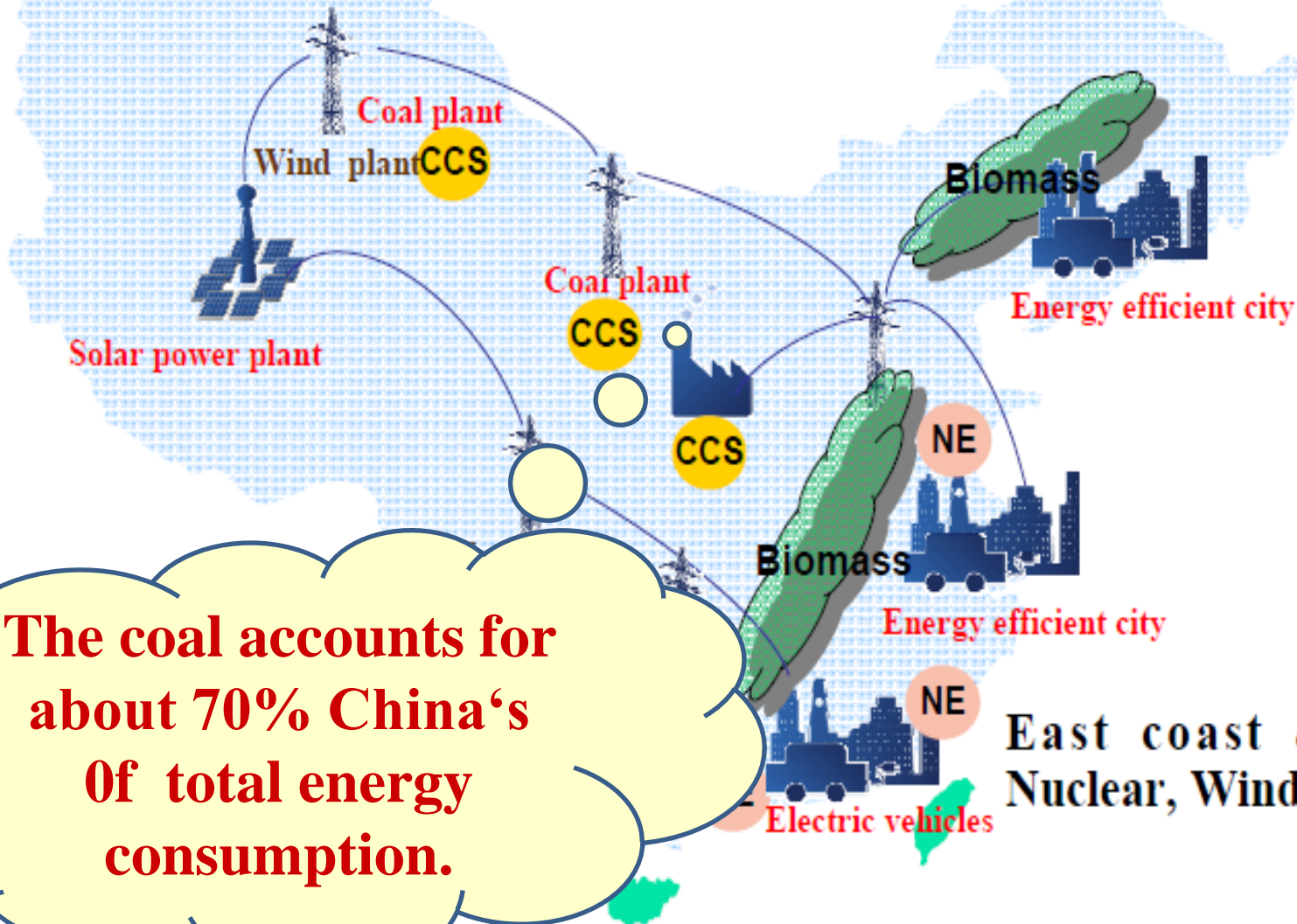
📖 Analysis and forecast on national electric power in China: In 2030, the electricity demand of per person will be about 2KW, total generation capacity will reach about 3000GW, the MW - level power stations will need 3000.











Localized Energy Resource in China

Northwest & Yellow River
Coal, Solar & Windy Energy

Northeast & South middle
Coal & Biomass Energy



List of countries by 2011 emissions estimates

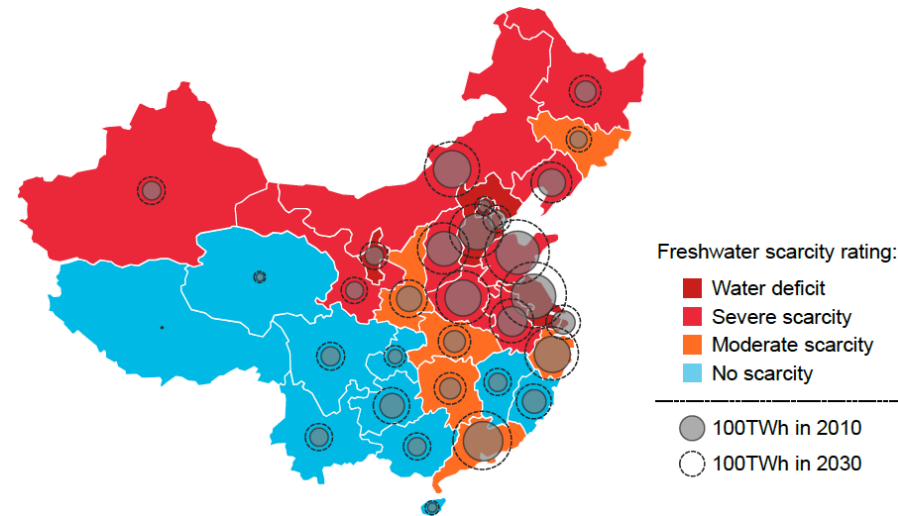
	Country	CO ₂ emissions	Area (in km ²)	Population	Emission / Person
	World	33,376,327	148,940,000	6,852,472,823	4.9
	China	9,700,000	9,640,821	1,325,852	7.2
	United States	5,420,000			17.3
	India	1,900,000			2
	Russia	1,850,000			12.8
	Japan	1,240,000			9.8
	Germany	810,000		81,799,600	9.9
	South Korea	610,000	100,210	48,875,000	12.6
	Canada	560,000	9,984,670	34,685,000	16.2

**China's
Environment
Challenge**

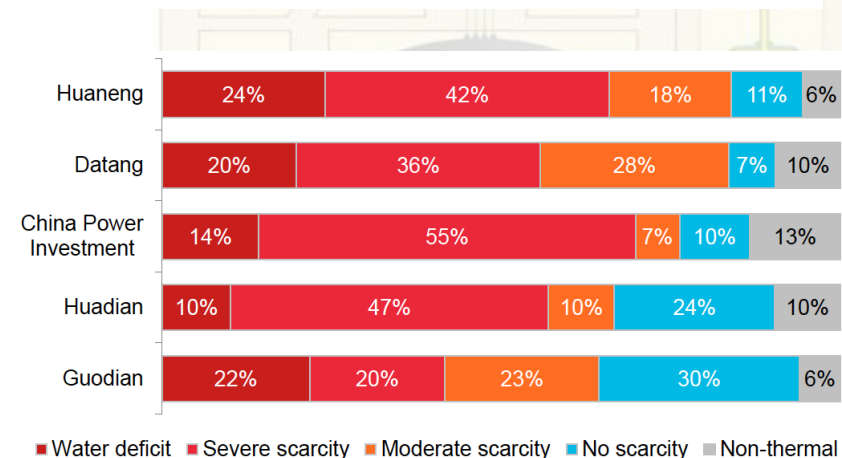
EDGAR (database created by [European Commission](#) and [Netherlands Environmental Assessment Agency](#)) released 2011 estimates. The following table lists the 2011 estimate of annual CO₂ emissions estimates (in thousands of CO₂ tonnes) from these estimates along with a list of emissions per capita (in tonnes of CO₂ per year) from same source.

Water Scarcity Threats China's Thermal Power Plants

- 📖 China's "Big Five" power utilities exposed to water disruption in their 500GW of plants;
- 📖 Their thermal plants used 102bn m³ of water for cooling in 2010 and forecasted to use even more in 2030;
- 📖 Most thermal (Coal) plants in China are located in water deficit region;
- 📖 Retrofitted to air-cooling uses less water but decreases efficiency by 3-10% and costs \$20bn per 100GW to current plants;
- 📖 Adopting non-water cooled power generation technology is a viable solution.



Power generation vs. water scarcity by province in China



China Big Five's capacity in water scarce region

Beyond Oil and Gas: The Methanol Economy**

George A. Olah*

Keywords:
 environmental chemistry · hydrocarbons ·
 hydrogen · methanol

Anthropogenic Chemical Carbon Cycle for a Sustainable Future

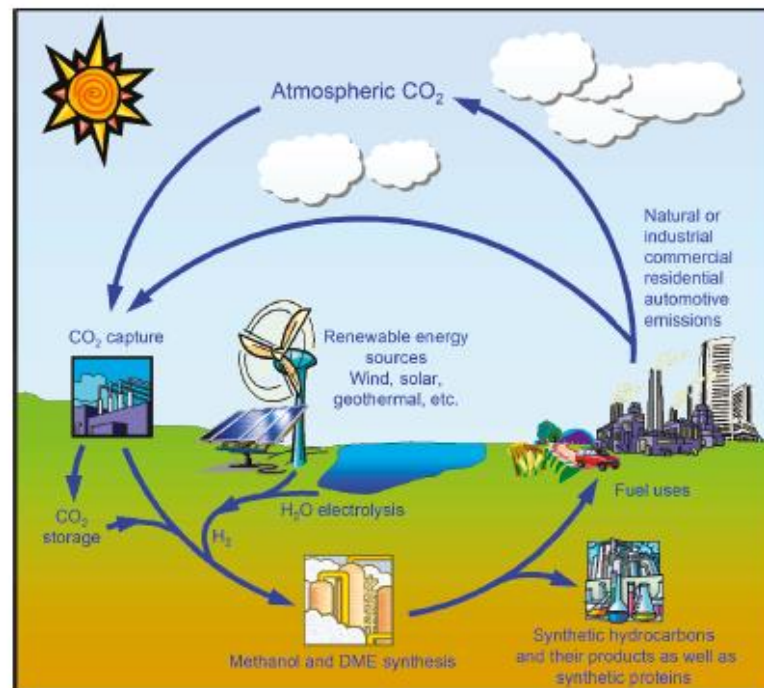
George A. Olah,* G. K. Surya Prakash, and Alain Goeppert

Loker Hydrocarbon Research Institute and Department of Chemistry, University of Southern California, University Park, Los Angeles, California 90089-1661, United States

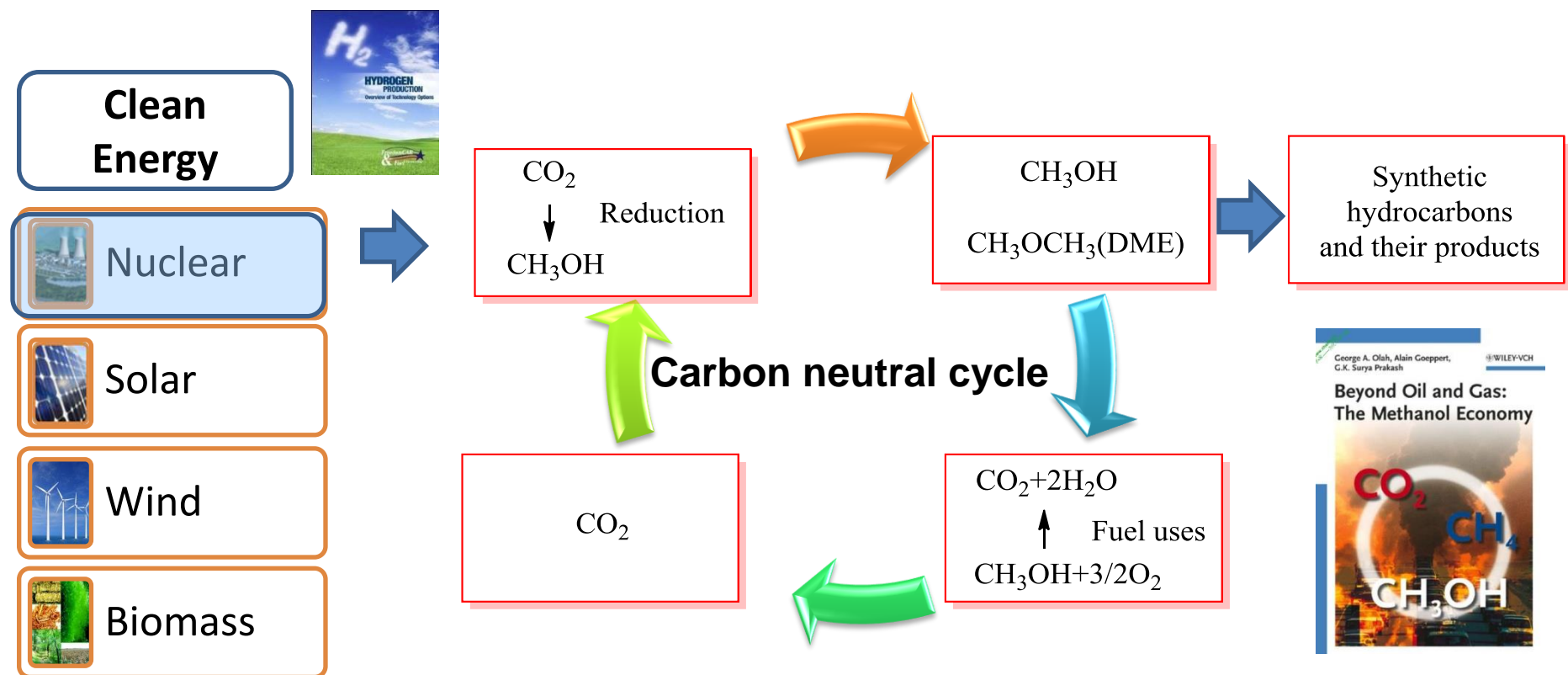
Hydrogen Production Using Nuclear Energy and Carbon Dioxide Hydrogenation to Methanol



On May 23, 2009, a pilot plant at the Mitsui Chemicals Osaka Works became the first site in the world to synthesize methanol from its carbon dioxide (CO₂) exhaust.



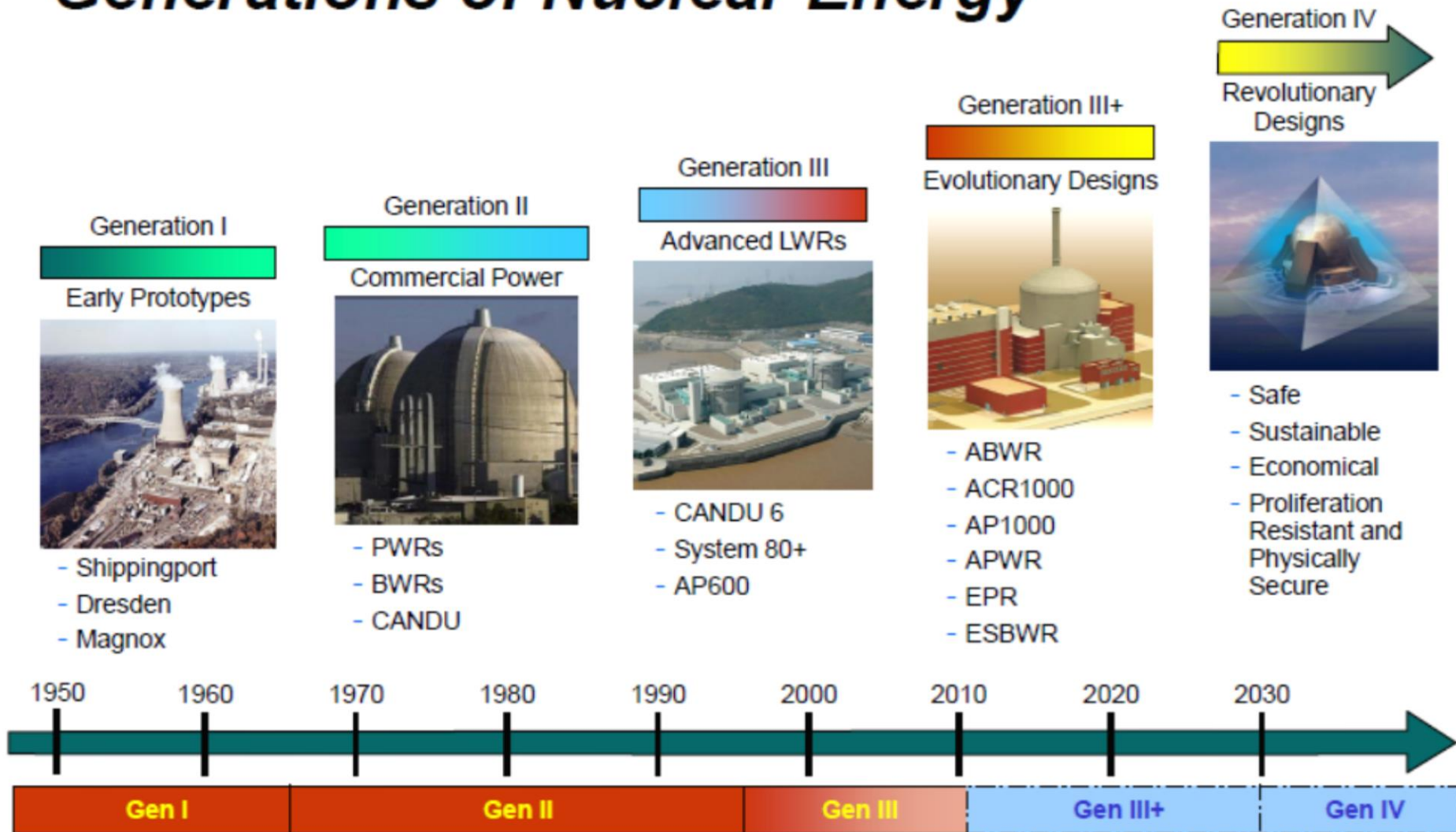
Methanol Economy: Carbon Neutral Cycle



George Andrew Olah
1994 Nobel Prize in Chemistry

1. Carbon neutral cycle based on methanol is one way to solve energy and environment problems.
2. Methanol producing from CO_2 and H_2 is the optional technology.
3. Hydrogen does not exist naturally, should get from clean energy without CO_2 emission.
4. The efficiency of methanol production from CO_2 should be improved.

Generations of Nuclear Energy



Overview of the Generation IV Systems

System	Neutron Spectrum	Fuel Cycle	Size (MWe)	Applications	R&D Needed
<i>Very-High-Temperature Reactor (VHTR)</i>	Thermal	Open	250	Electricity, Hydrogen, Process Heat	Fuels, Materials, H ₂ production
<i>Supercritical-Water Reactor (SCWR)</i>	Thermal, Fast	Open, Closed	1500	Electricity	Materials, Thermal-hydraulics
<i>Gas-Cooled Fast Reactor (GFR)</i>	Fast	Closed	200-1200	Electricity, Hydrogen, Actinide Management	Fuels, Materials, Thermal-hydraulics
<i>Lead-Cooled Fast Reactor (LFR)</i>	Fast	Closed	50-150 300-600 1200	Electricity, Hydrogen Production	Fuels, Materials
<i>Sodium Cooled Fast Reactor (SFR)</i>	Fast	Closed	300-1500	Electricity, Actinide Management	Advanced recycle options, Fuels
<i>Molten Salt Reactor (MSR)</i>	Epithermal	Closed	1000	Electricity, Hydrogen Production, Actinide Management	Fuel treatment, Materials, Reliability

Why G IV

Safe:

- **Residual heat** will be removed with active cooling system in GIII ;
- **Residual heat** will be removed with passive cooling system in GIV (natural cycle), aiming to inherent safe.

Substanable:

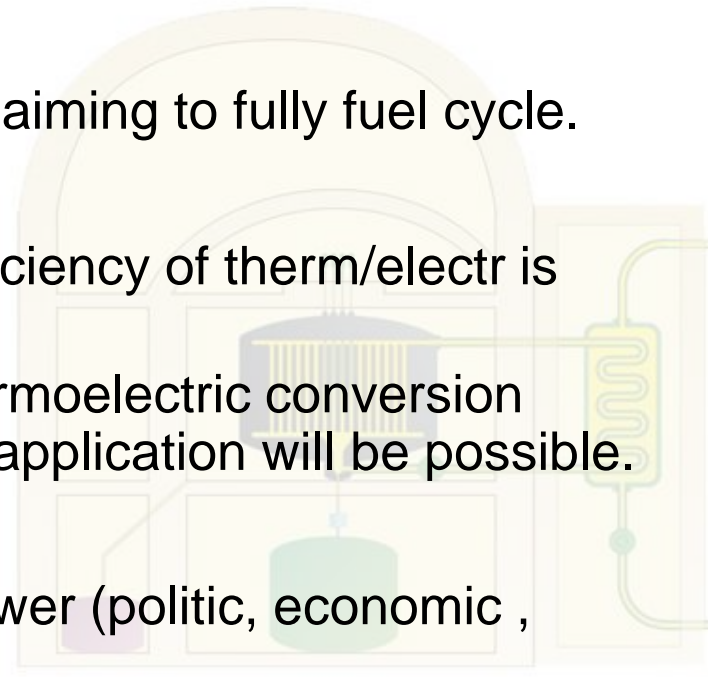
- Fuel availability is 1~2% in GIII;
- Fuel availability is large than 10% in GIV, aiming to fully fuel cycle.

Economic:

- Output heat is about 300°C in GIII , Efficiency of therm/electr is about 33%;
- Output heat is > 500°C in GIV , the thermoelectric conversion efficiency will be higher, and the hybrid application will be possible.

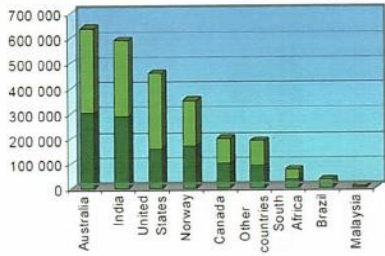
Non-proliferation:

- The proliferation is controlled by man-power (politic, economic , military) in GIII;
- Proliferation resistance is physical in GIV.



About Thorium

India's thorium reserves stimulate its thorium power development.



India's nuclear strategy

1. Heavy water reactors for unenriched, limited uranium reserves.
2. Fast breeder reactor for plutonium from spent fuel uranium
3. Thorium fast breeder reactor.

India has 13 heavy water reactors plus 4 under construction.

The CANDU-like technology allows breeding U-238 to Pu-239 and Th-232 to U-233.

India already has reprocessing facilities and a developmental breeder reactor.

Kamini reactor tests U-233 from Kalpakkam experimental breeder.

0.5 GW fast breeder reactor is under construction, due 2010.

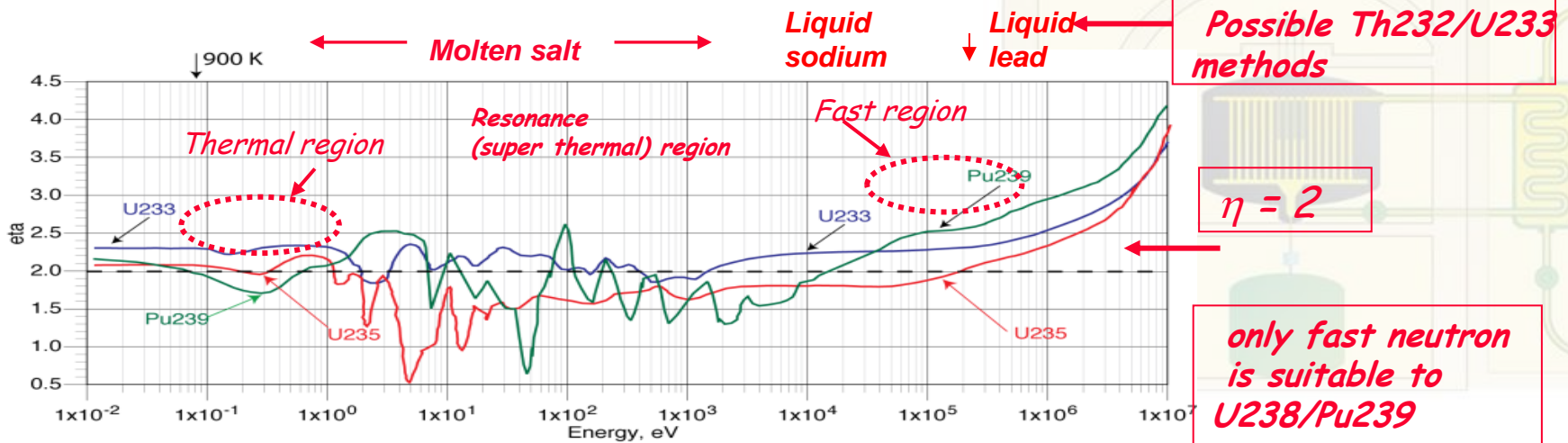
20 GW of U and Th power by 2020.
30% of electricity from Th by 2050.

India has little uranium, which has been difficult to obtain, because India did not sign the Nuclear Non-proliferation Treaty.

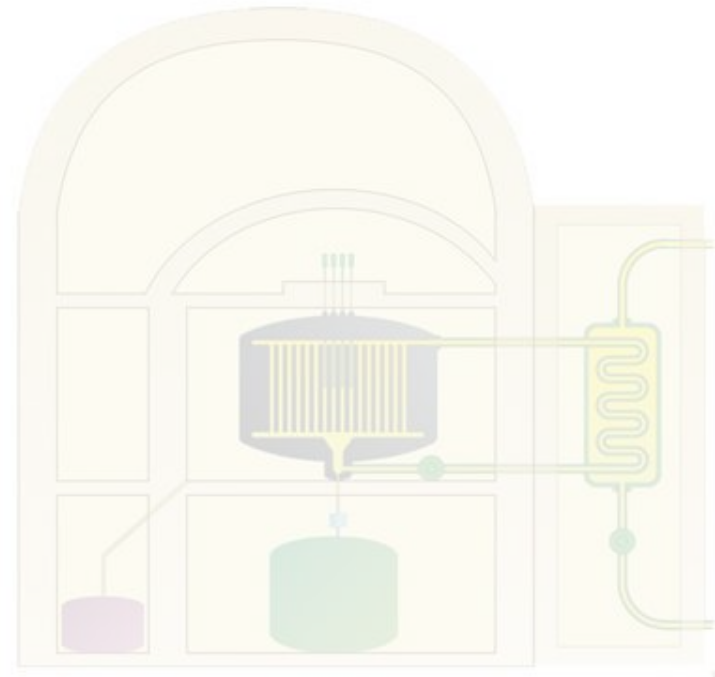
India's thorium power reactor development uses solid fuel, not liquid fluoride salts as the LFTR uses.

CONCLUSION –C.Rubia

Unlike other energy sources, China's reserves of Thorium, may ensure the major domestic energetic supply for many centuries to come. For instance the whole China's today electricity (3.2 Trillion kWh/year) could be produced during $\approx 20,000$ years by well optimized Th reactors and 8,9 million ton of Th, a by-product of the China's REE basic reserves.

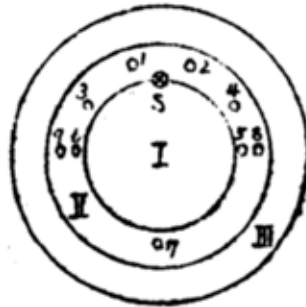
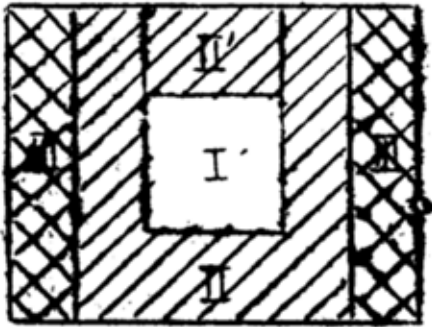


TMSR



Early Efforts for MSR in China

1970 - 1971, SINAP built a zero-power (cold) MSR.



I - core
II - reflector
II' - reflector cover
III - protection wall
S - neutron source
(100mCi Ra-Be)





1-2- safety rod
3- regulating rod
4- shim rod
5-6- backup safety rod
7-8-9- BF₃ neutron counter

1972 - 1973, SINAP built a zero-power LWR.

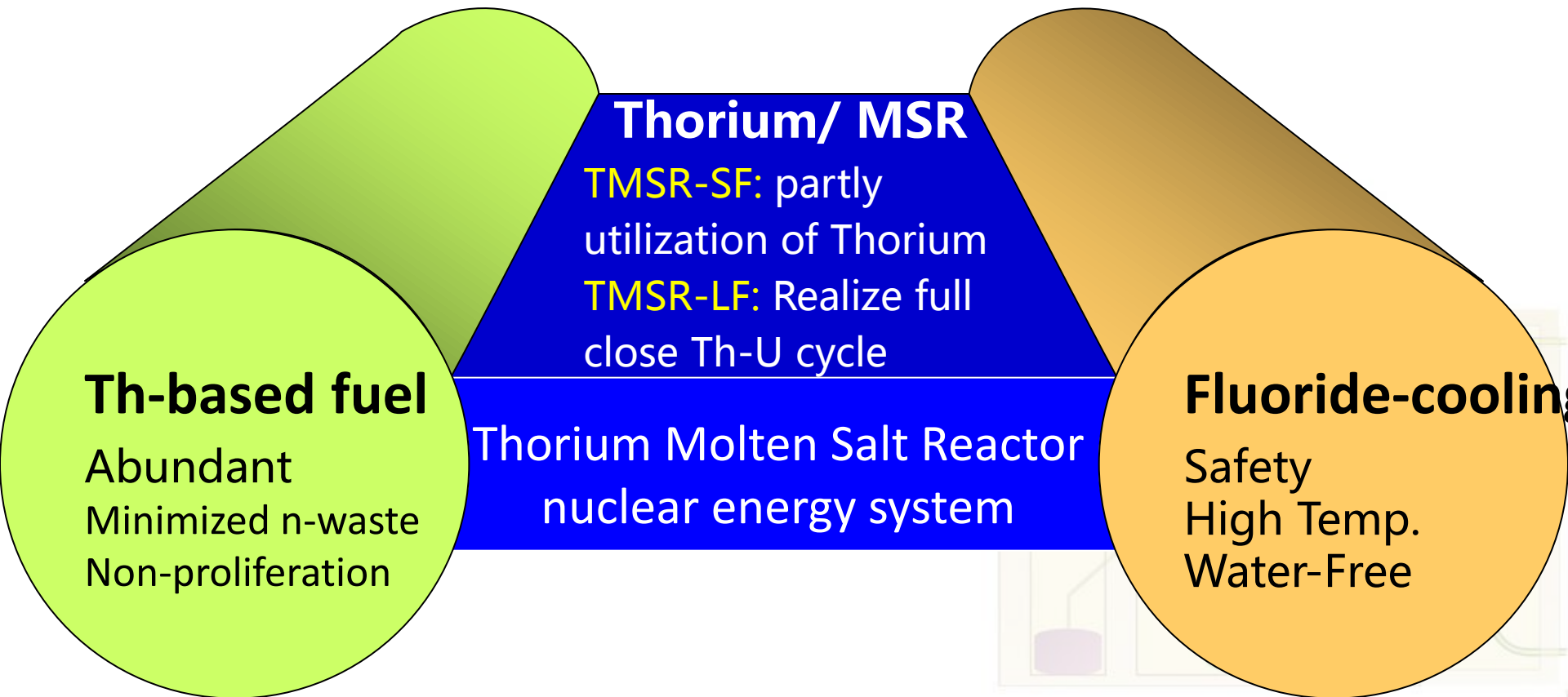


1972-1975, in SINAP about 400 scientists and engineers designed the Qinshan 300 MWe (Qinshan NPP-I), which has been operating since 1991.

Thorium Molten Salt Reactor Nuclear Energy System - TMSR

-  The Aims of TMSR is to develop Th-Energy, Non-electric application of Nuclear Energy based on TMSR(LF) and TMSR(SF) during coming 20-30 years.
-  The program initiated by CAS from 2011
-  TMSR(LF) 液态燃料钍基熔盐堆 --- MSRs
-  TMSR(RF) 固态燃料钍基熔盐堆 --- FHR s

Thorium Molten Salt Reactor nuclear energy system



Outline

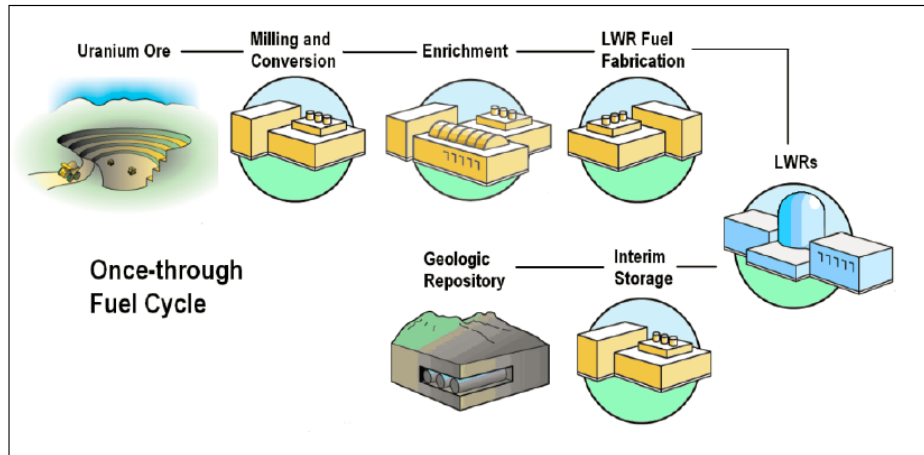
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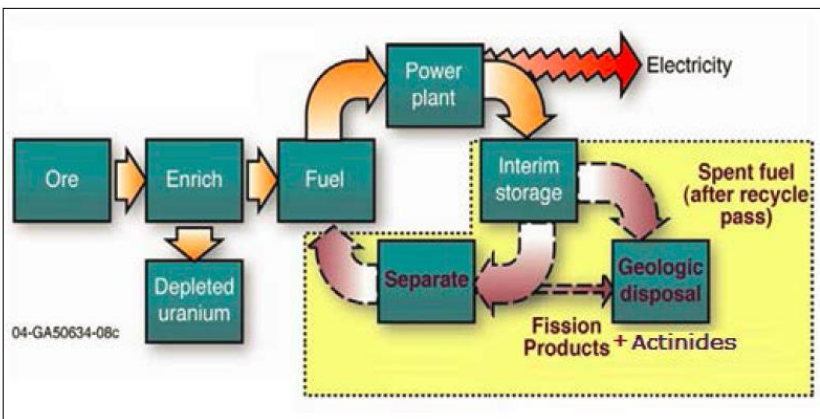
Nuclear Fuel Cycle Modes



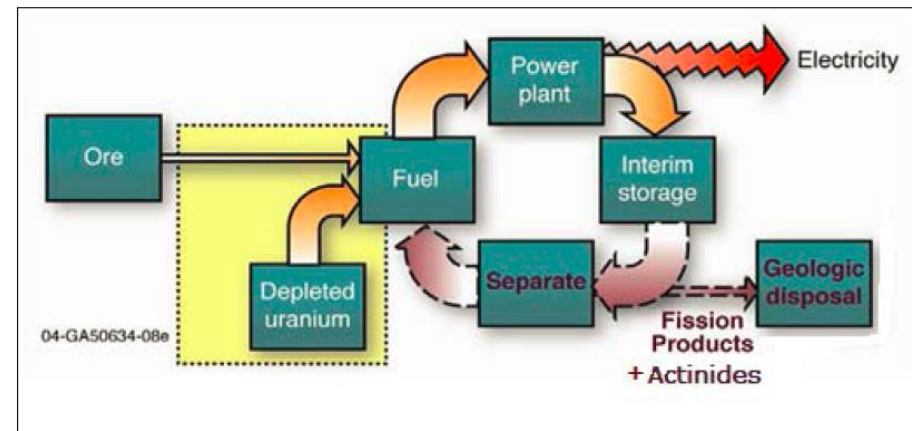
The Once-Through Fuel Cycle



Alternate Once-Through Fuel Cycles

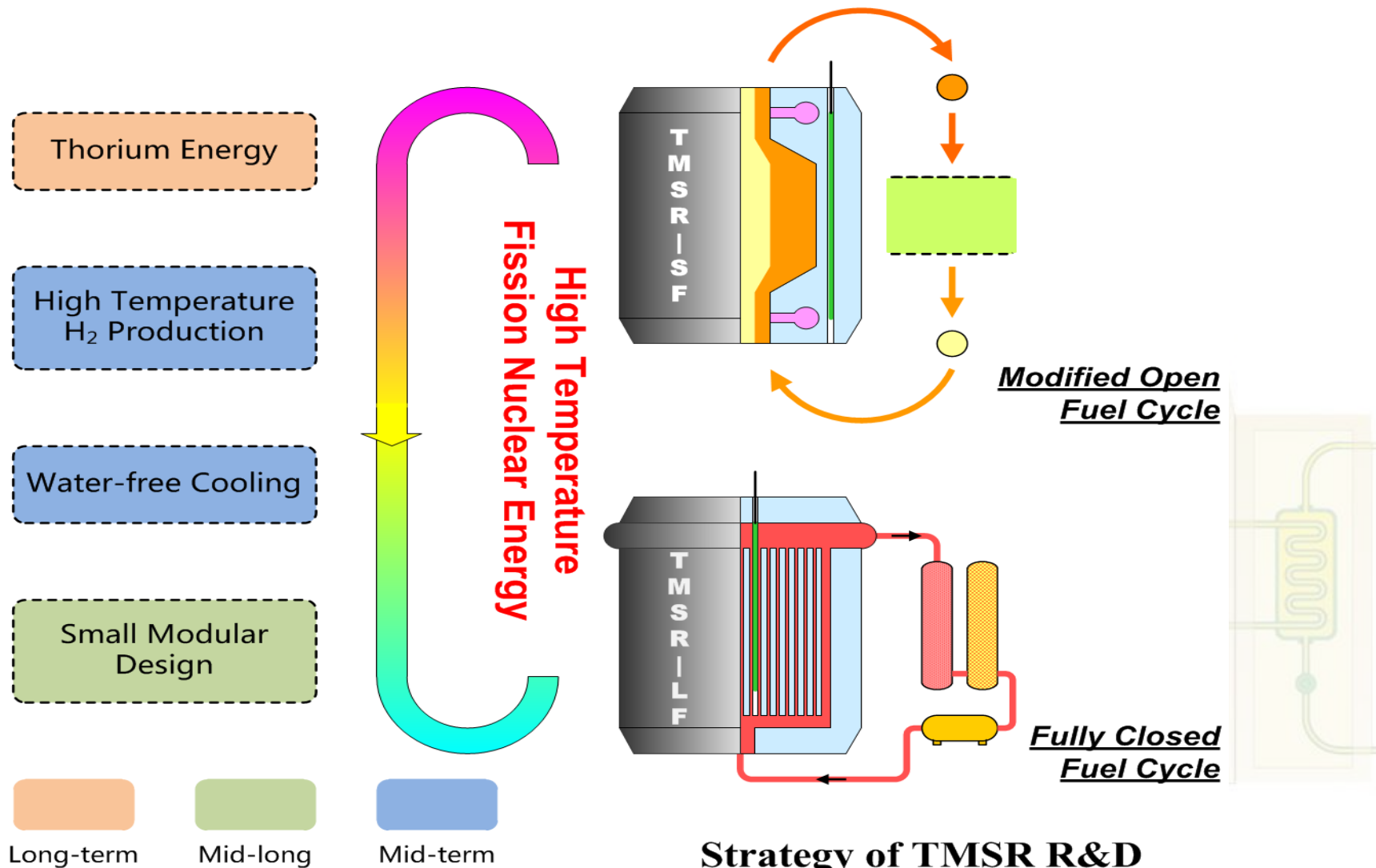


Modified Open Fuel Cycle

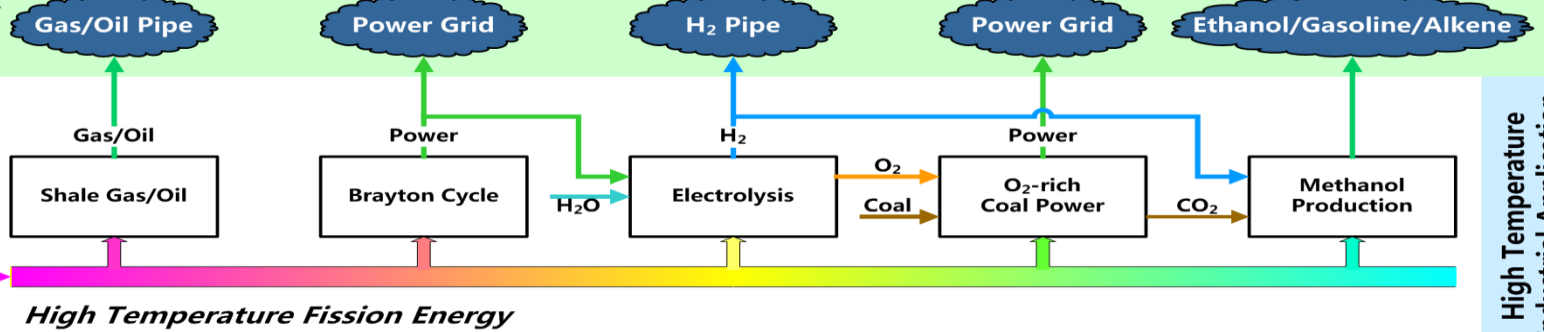


***Fully-Closed Fuel Cycle
with (Sustained) Recycle***

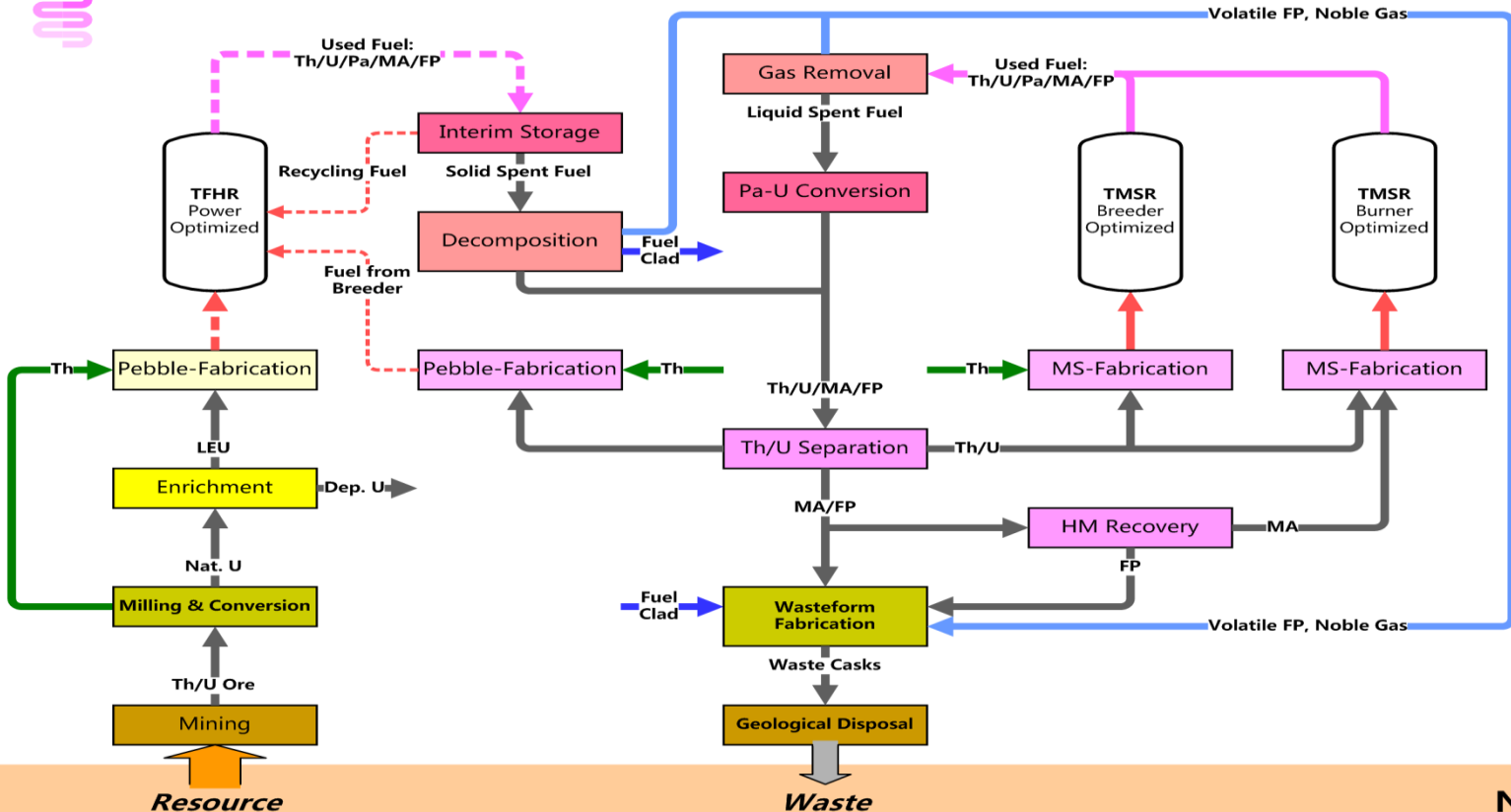
Reactors and Applications



SOCIETY




High Temperature Industrial Application




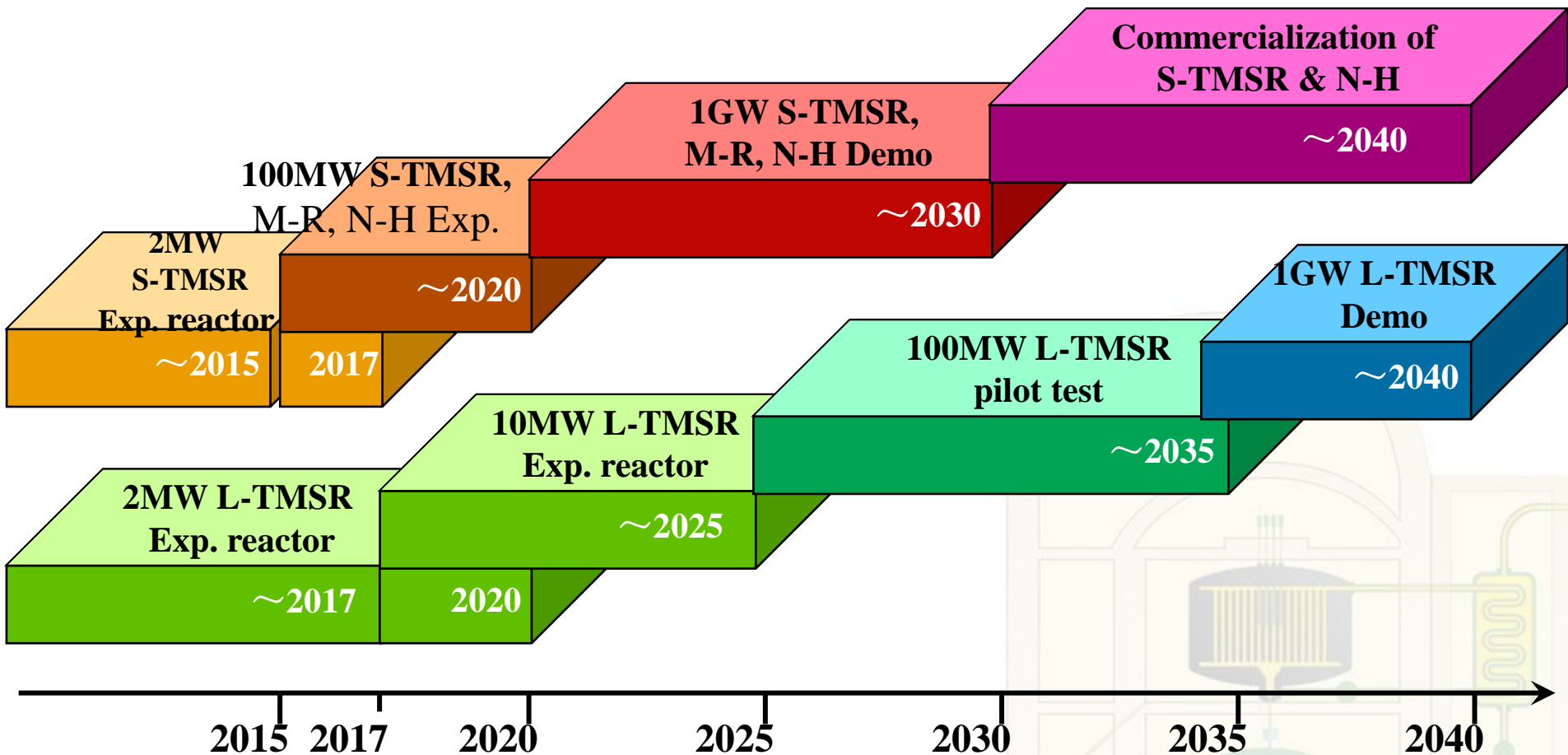
Thorium Based Reactor System

NATURE

FHRs Can Be Considered A Precursor to MSRs

 MSRs development require all of the technologies required for an FHR (such as: materials, pumps, heat exchangers, and salt chemistry and purification, and power conversion) except coated particle fuel

 FHRs deployment does not require some of the MSR longer-term development activities (such as , reprocessing of highly radioactive fuel salts), can be deployed much earlier than MSRs



Th-U cycle

MS coolant

Nuclear power plant

Hydrogen production with high temperature nuclear power

2012/03/12

Xiaohan Yu

TMSR WBS (0)

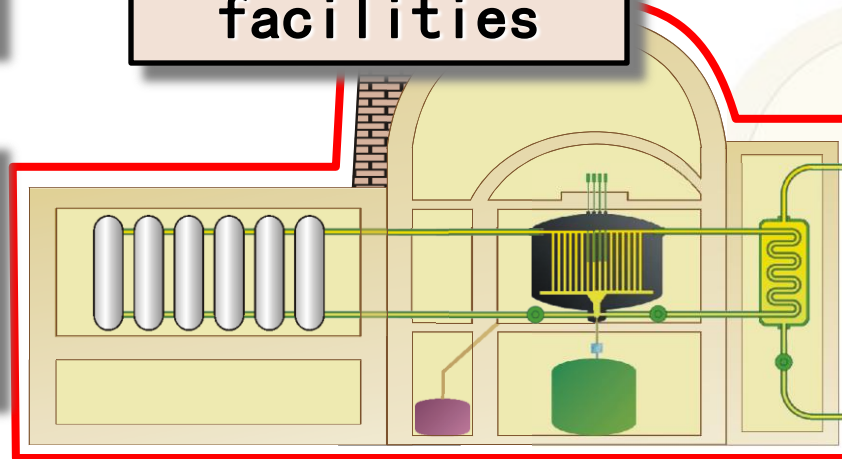
1. Reactor Phys.
&Eng.

2. Molten-Salt
Chem. &Eng.

3. Th-U
Radioactive
Chem. &Eng.

4. Nuclear Power
Material & Eng.

0. supporting
facilities



5. Th-U Security &
Eng.

Reactor
Modeling

Thermal-
Power Cycle

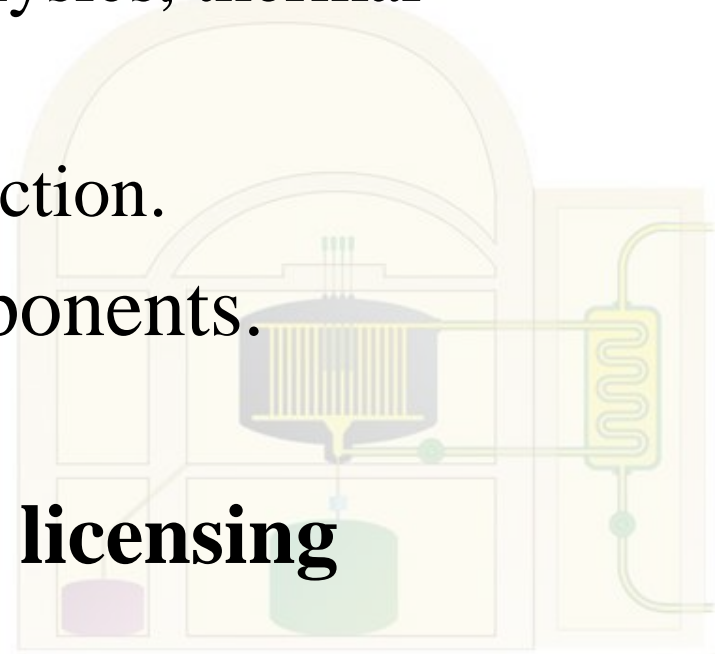
Nuclear
Hydrogen &
CO₂/Carbon
Utilization

R&D abilities –I for TMSR deveopment

TMSR reactor design and development.

- ❑ Reactor core design: neutron physics, thermal hydraulics...
- ❑ Engineering design and construction.
- ❑ Key technologies and components.

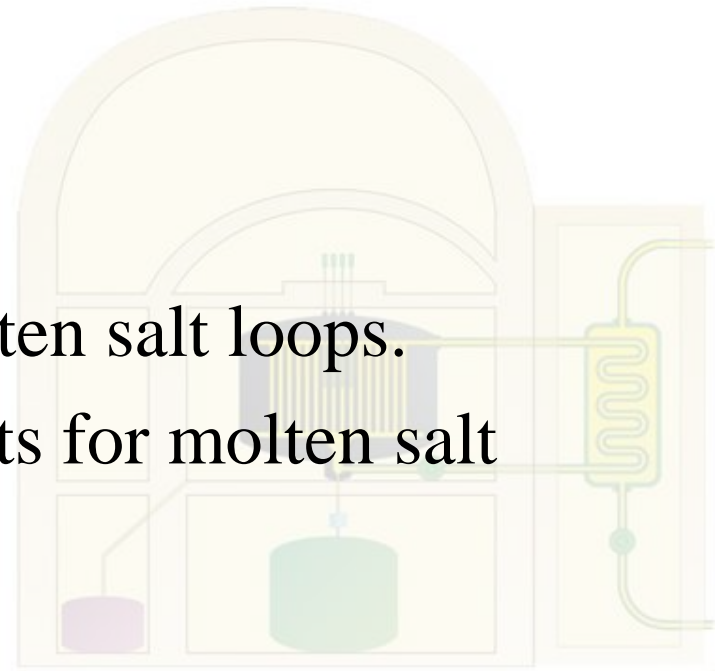
Developing safety codes and licensing



R&D abilities -II for TMSR deveopment

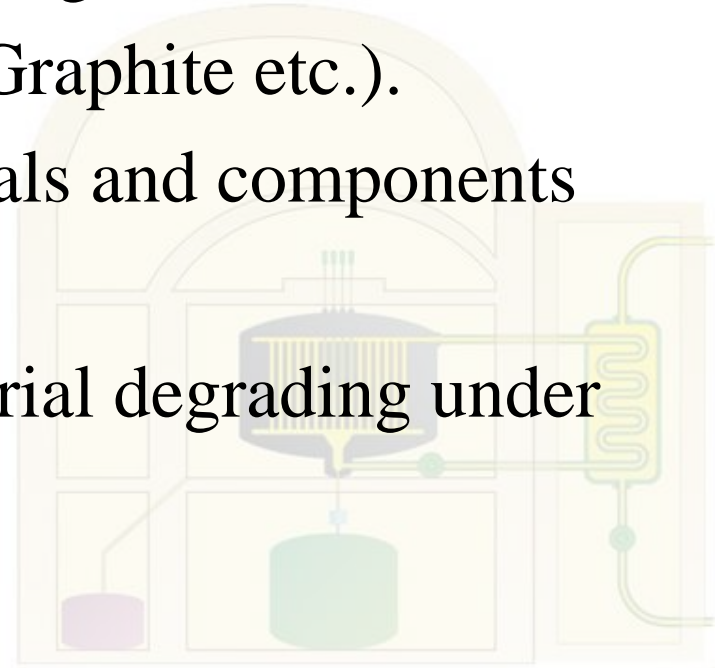
Molten salt manufacturing techniques and molten salt loop techniques

- ❑ Separation of ^7Li .
- ❑ Purification of fluoride salt.
- ❑ Design and construction of molten salt loops.
- ❑ Development of key components for molten salt loop.



Materials for TMSRs

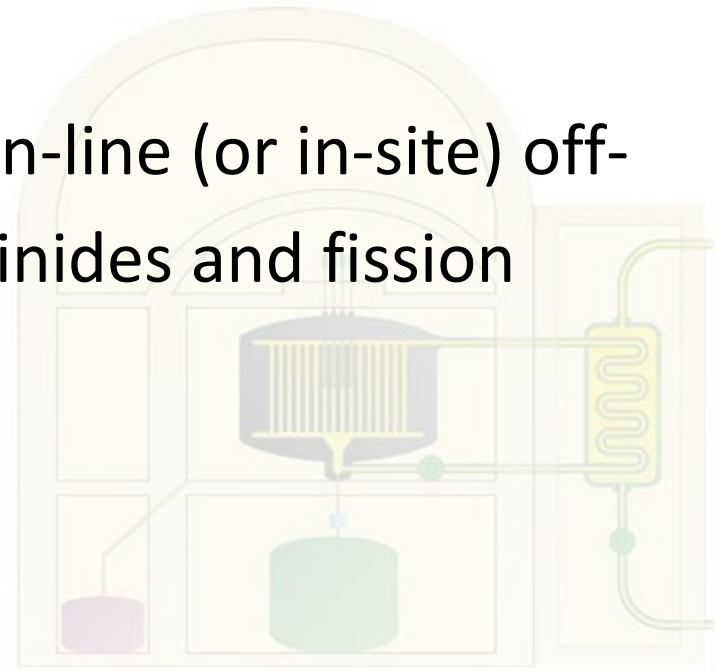
- ❑ Production, processing and testing of structure materials TMSR (Hastalloy-N, Graphite etc.).
- ❑ Carbon-based structure materials and components for TMSR
- ❑ Effect and mechanism of material degrading under service condition.








Th/U fuel and Pyro-process techniques

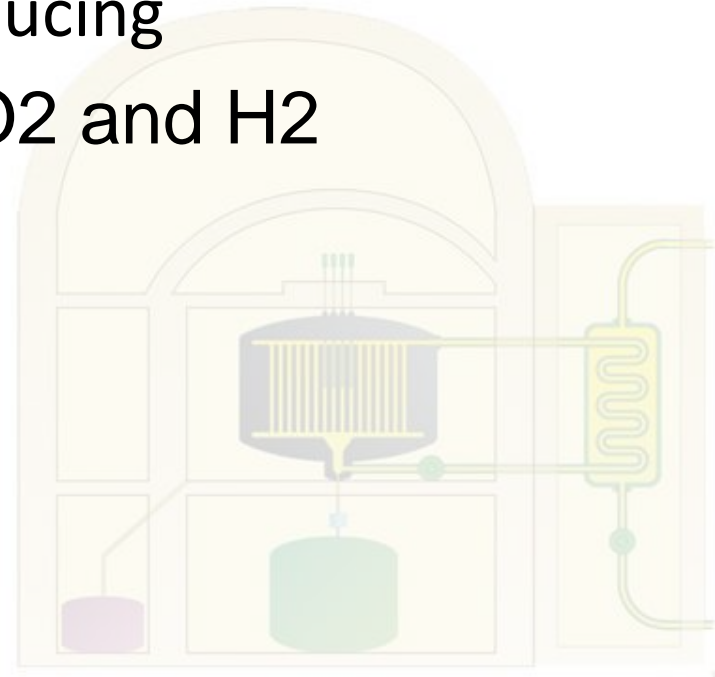
- ❑ Production of nuclear-grade Thorium fuel (both fluoride and oxide).
- ❑ Triso fuel manufacturing
- ❑ Pyro-process techniques for on-line (or in-site) off-line) chemical separation of actinides and fission product for Th/U fuel cycle).



Techniques of Nuclear Hybrid Energy system

 100kW CSP (the salt is absorber and heat storage media)

-  Techniques for hydrogen producing
-  Methanol producing from CO₂ and H₂
-  CSP techniques



Outline

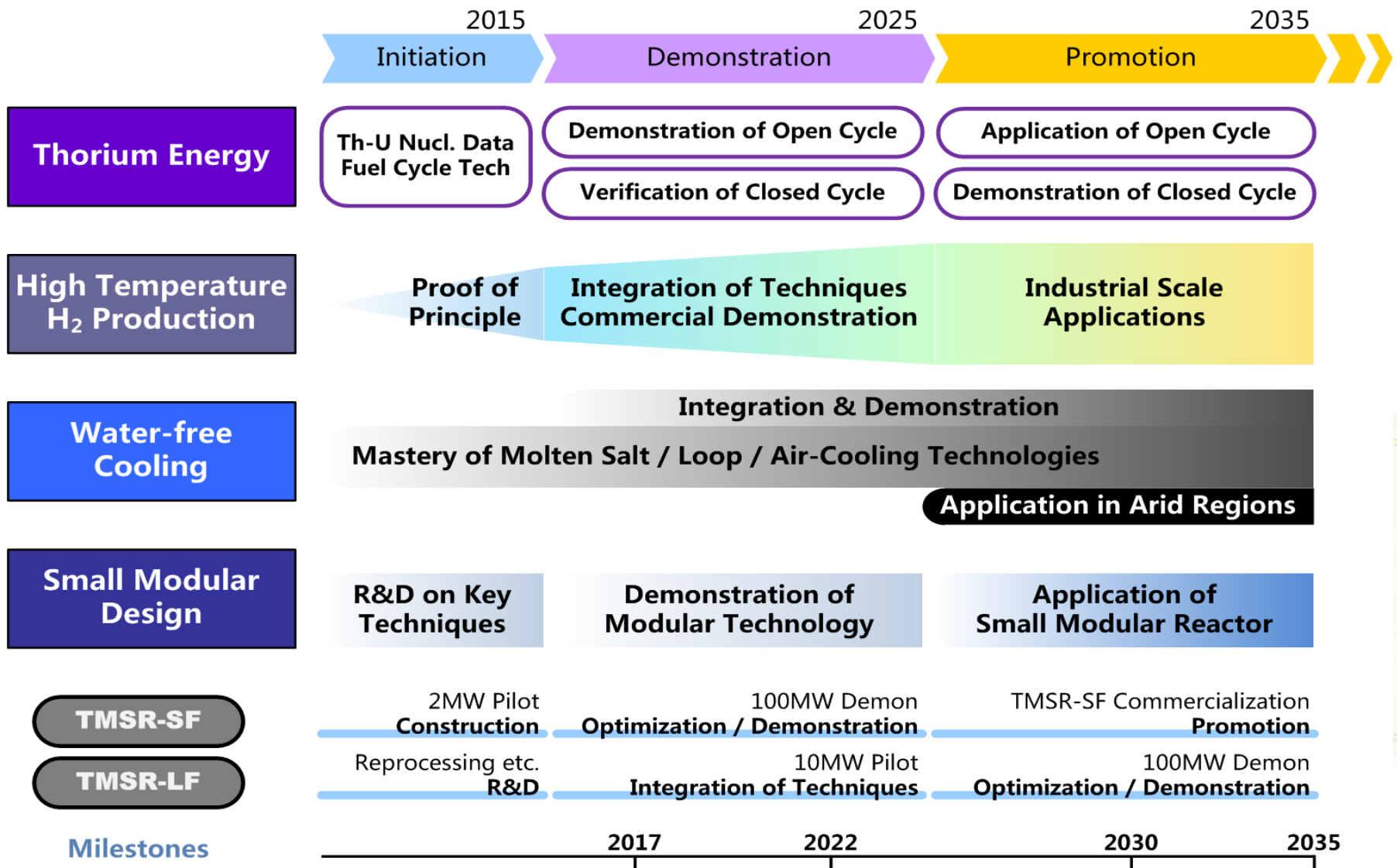
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TMSR Schedules

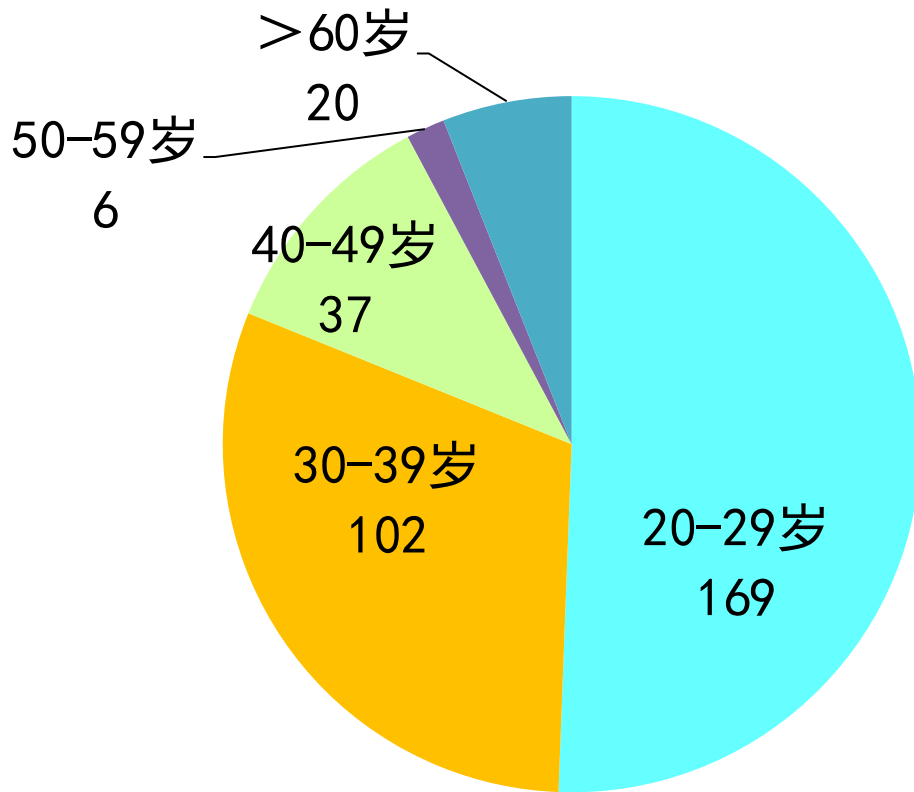


Team and Organization

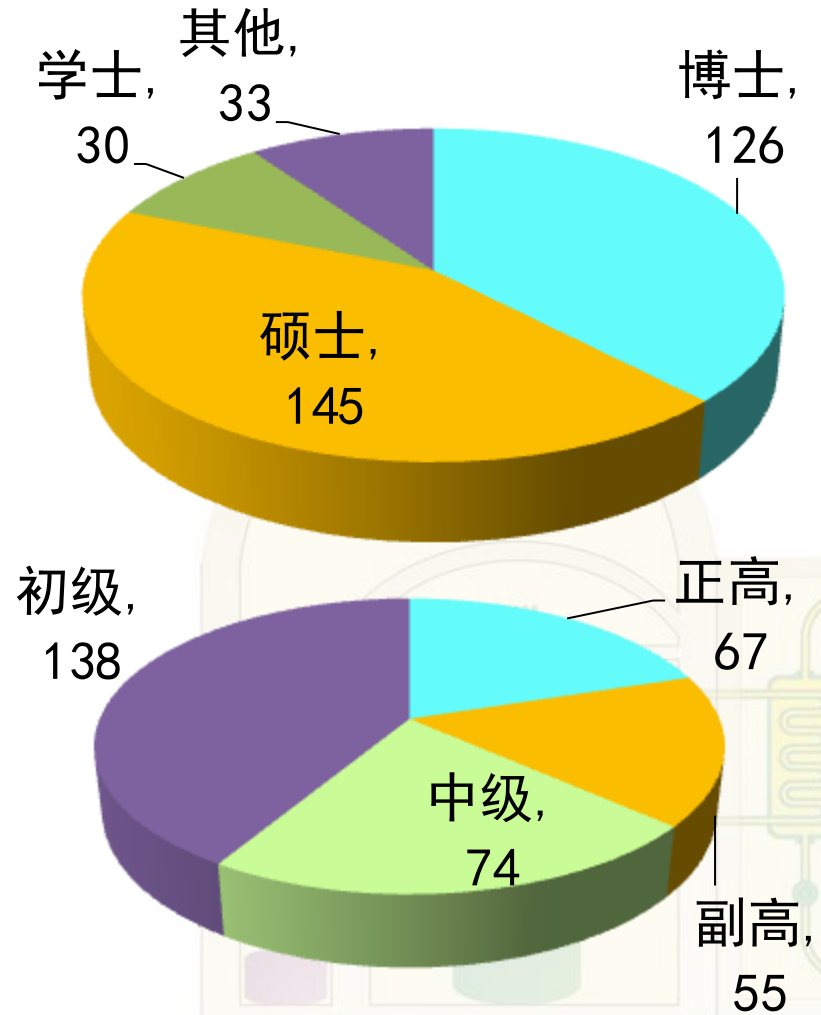
- 📖 Basic team –from SINAP (some have the experiences of SSRF construction). It take time for them to be professional researchers
- 📖 Young scientists and engineers are majority – for the future.
- 📖 To attract the exellent younger and experienced scientists both from domestic and abroad.



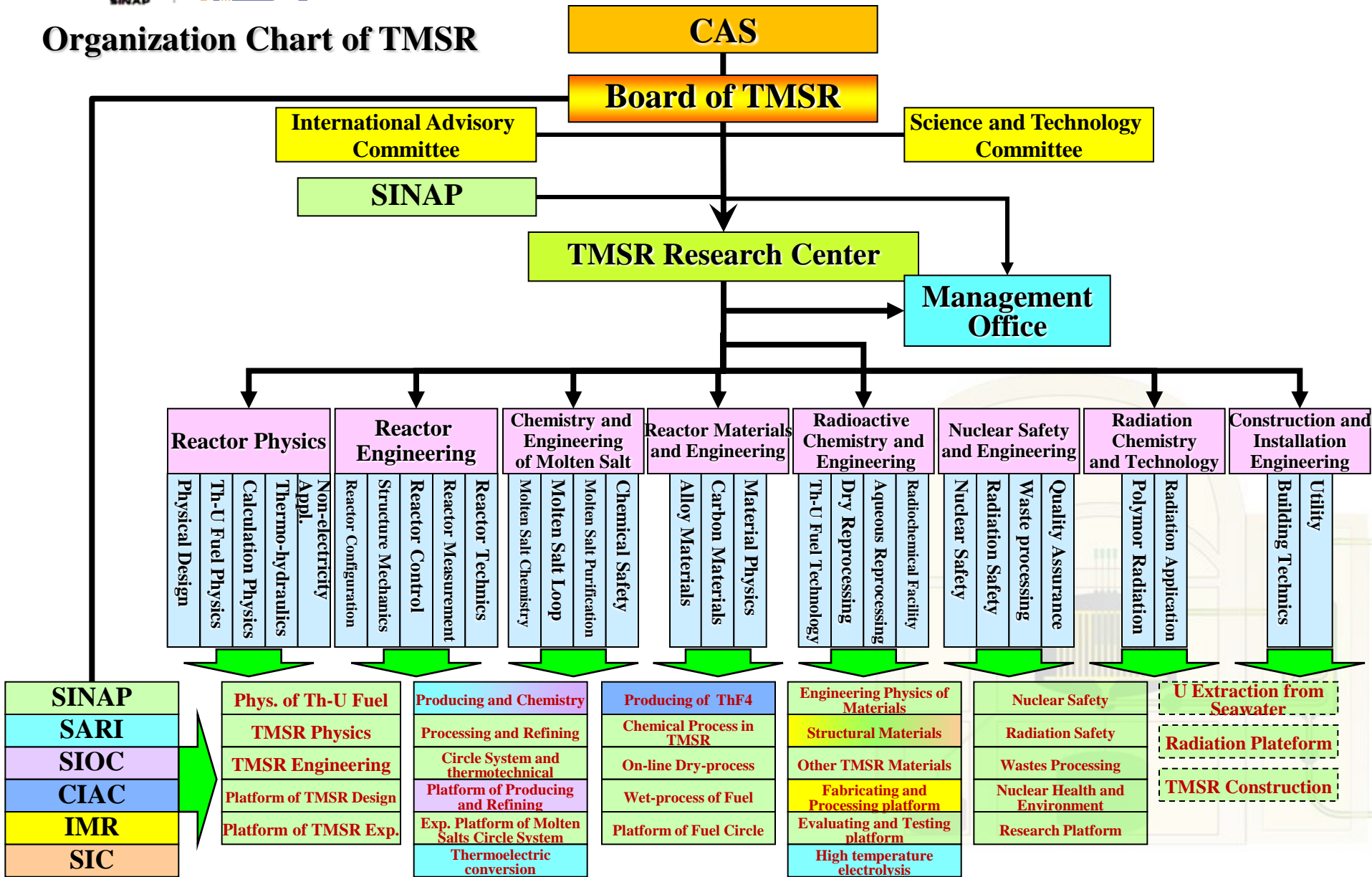
Team Structure of TMSR () 400)



- ◆ TMSR staff Average Age ~31
- ◆ Key personnel Average Age ~38



Organization Chart of TMSR

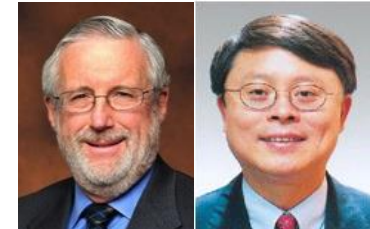


Organizational Overview

The Chinese Academy of Sciences (CAS) and U.S. Department of Energy (DOE) Nuclear Energy Cooperation Memorandum of Understanding (MOU)

MOU Executive Committee Co-Chairs

China – Mianheng Jiang (CAS) 江绵恒
U.S. – Peter Lyons (DOE)



Technical Coordination Co-Chairs

China – Zhiyuan Zhu (CAS) 朱志远
U.S. – Stephen Kung (DOE)



Nuclear Hybrid Energy Systems *

- Zhiyuan Zhu (CAS) 朱志远
- Yuhan Sun (SARI,CAS) 孙予罕
- Steven Aumeier (INL)

* Work scope governed by DOE-CAS Science Protocol Agreement

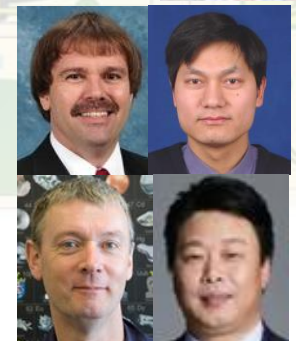
Molten Salt Coolant Systems

- Hongjie Xu (SINAP, CAS) 徐洪杰
- Weiguang Huang (SARI,CAS) 黄伟光
- Cecil Parks (ORNL)
- Charles Forsberg (MIT)

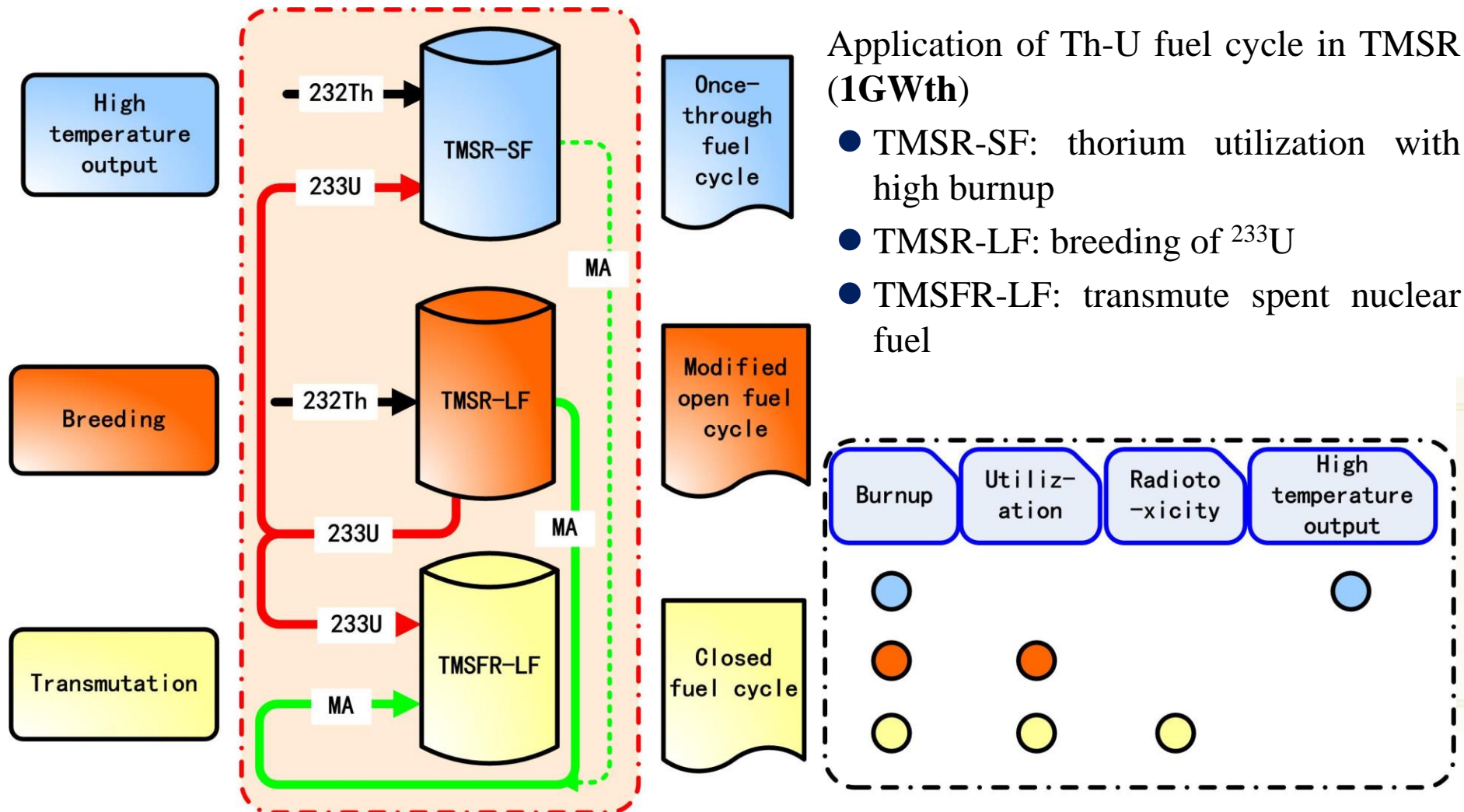


Nuclear Fuel Resources

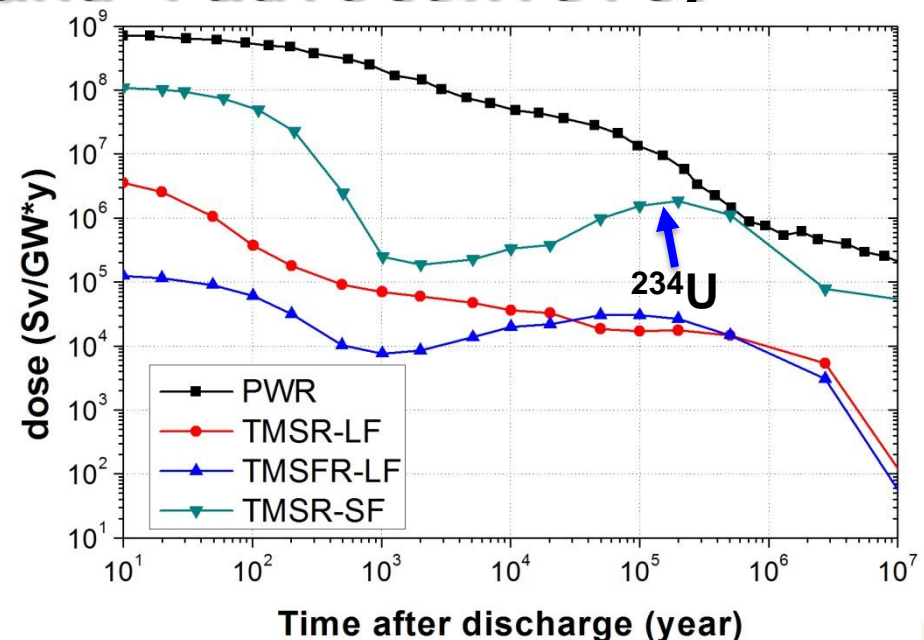
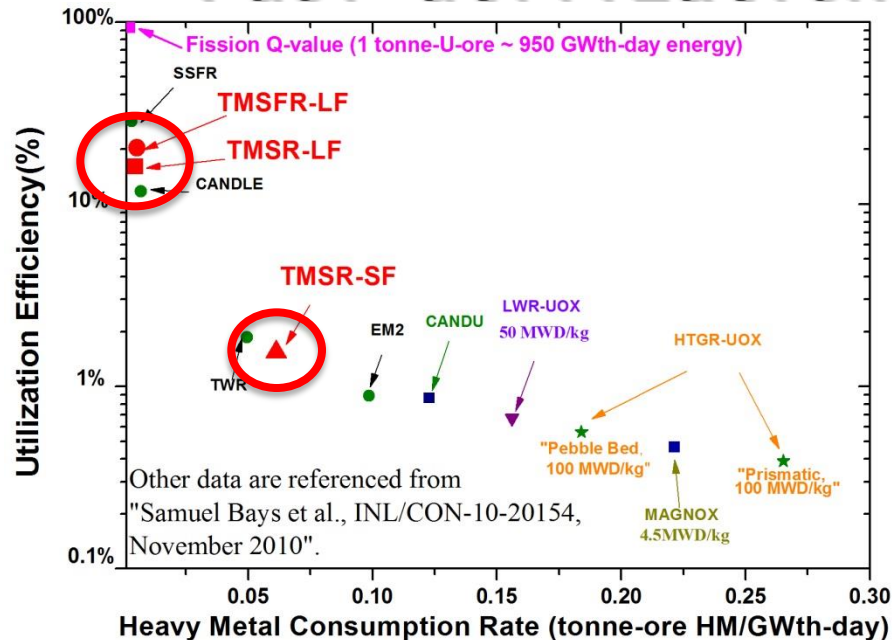
- Zhimin Dai (SINAP, CAS) 戴志敏
- Biao Jiang (SARI,CAS) 姜标
- Phil Britt (ORNL)
- John Arnold (UC-Berkeley)



I、Preliminary Physical Analysis of Th/U Fuel Cycle Based on MSR (1GWth)



Fuel utilization and radiotoxicity







TMSR-LF with online processing can significantly improve the fuel utilization.

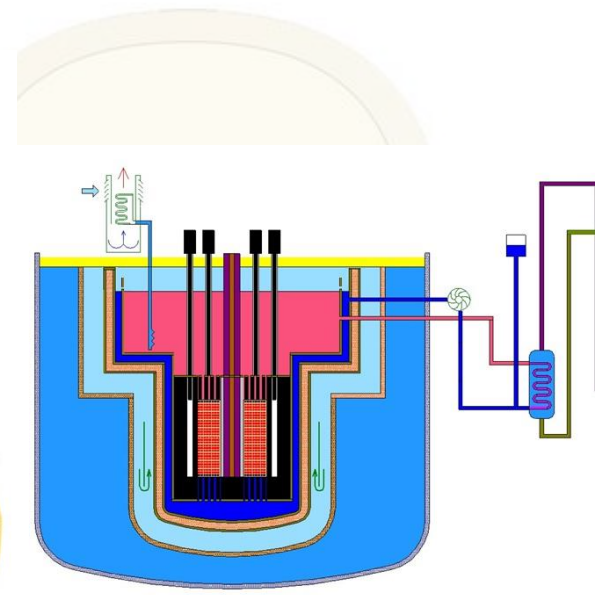
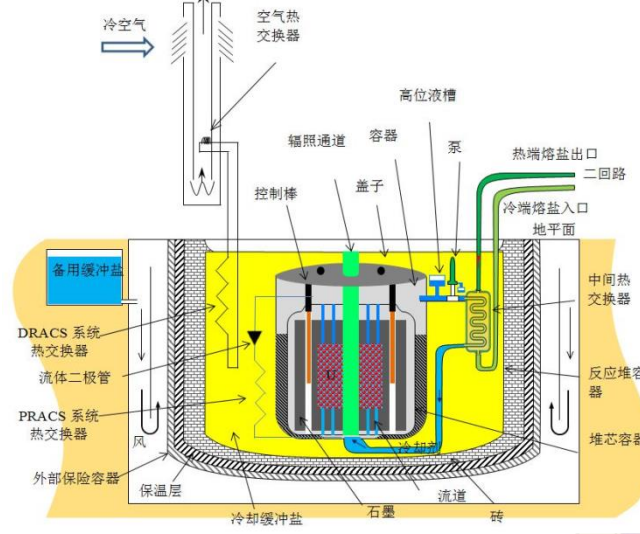
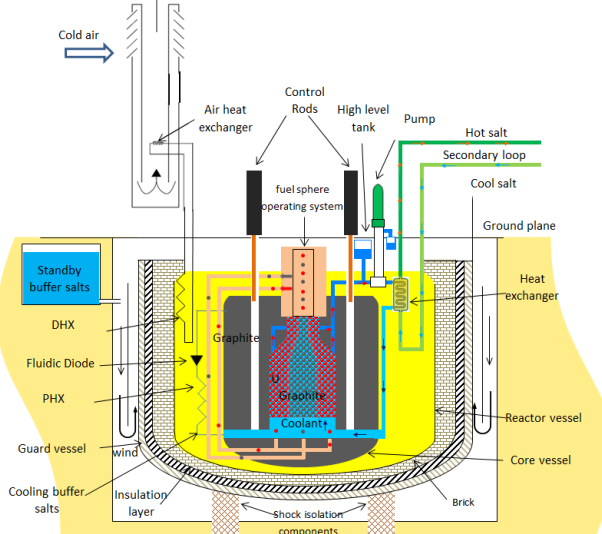
TMSR-SF has far higher radiotoxicity than TMSR-LF, but still lower than traditional PWR.

- Higher burnup of TMSR-SF to save uranium resource.
- TMSR-LF and TMSFR-LF to realize self-sustaining and produce U-233.
- TMSFR-LF is optimized for transmutation.

A Combination of TMSRs to realize self-sustaining Th-U fuel cycle and nuclear waste minimization.

II、Goals and physical design for TMSR-SF1

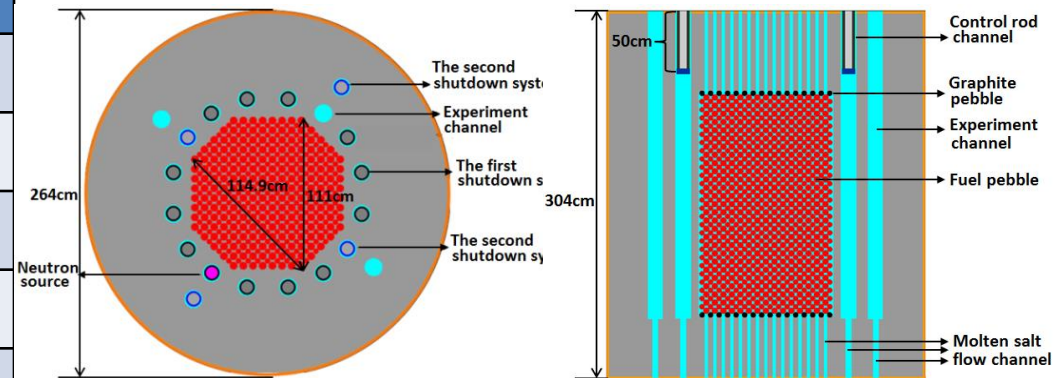
-  Form abilities for design: physics design, thermal-hydraulics design, safety system design and engineer design
-  Form abilities for integration, operation and maintenance.
-  Verify characteristics of FHR: physics behavior, safety behavior and thermal-hydraulics behavior
-  Research behavior of material and fuel



From partition flow pebble bed design to No-partition ordered pebble bed design

Core geometry of TMSR-SF1

Parameters	Value
Fuel Pebble diameter	6.0 cm
Number of Fuel Pebble	11043
Number of Graphite Pebble	432
Height of active core	185 cm
Side by side distance of active core	111cm 114.9 cm
Height of reflector	300.0 cm
Diameter of reflector	260.0 cm
Thickness of upper reflector	65.0 cm
Thickness of bottom reflector	50.0 cm
Thickness of side reflector	73.5 cm

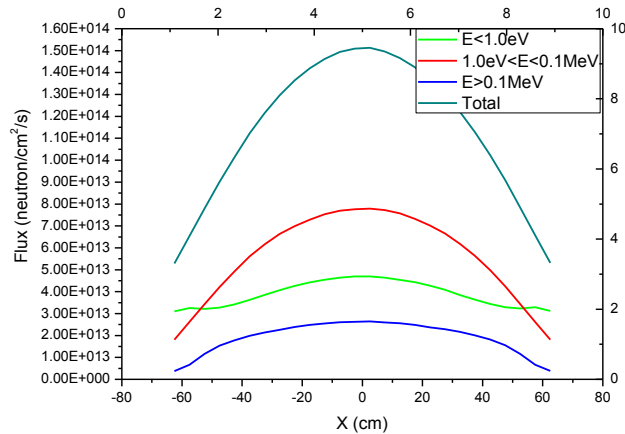


📖 Ordered-bed active core.

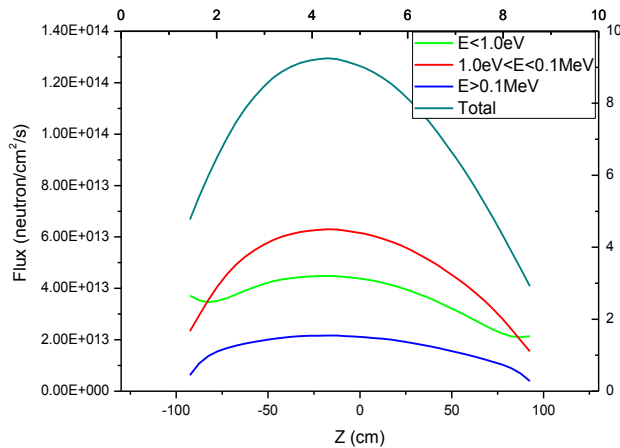
📖 Control rods: 12 regulating rods , 4 safety rods.

📖 Other channel:1 neutron source channel, 3 experimental channels.

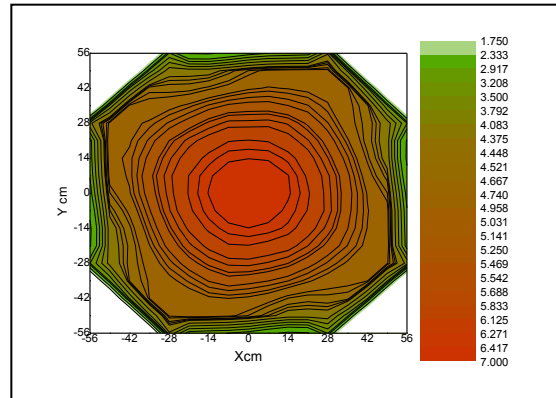
Neutron flux and power distribution



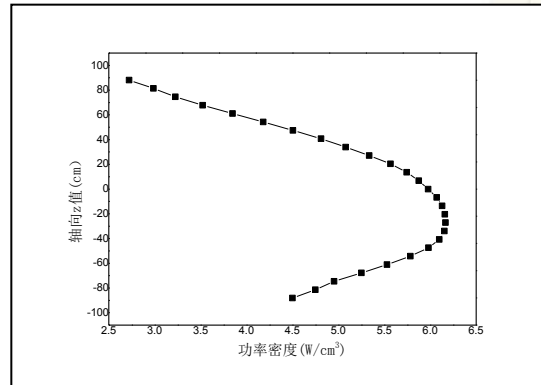
Radial neutron flux distribution



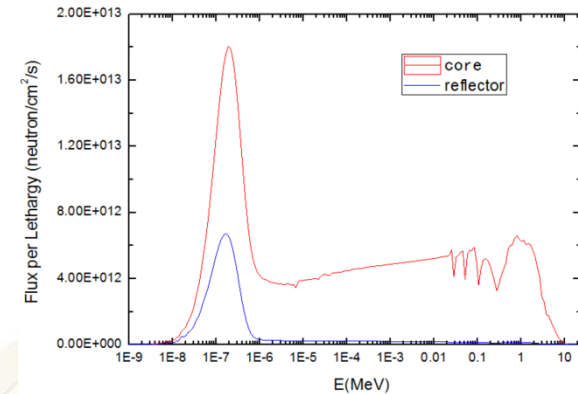
Axial neutron flux distribution






Horizontal power distribution



Axial power distribution



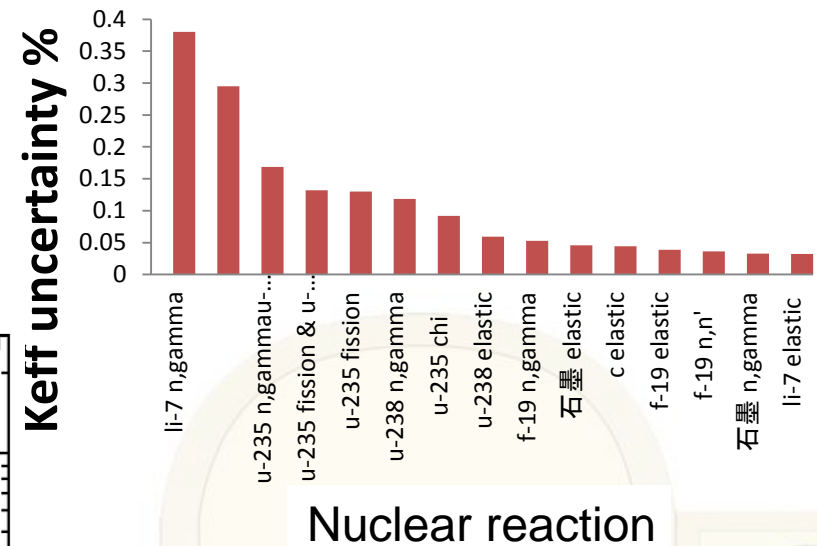
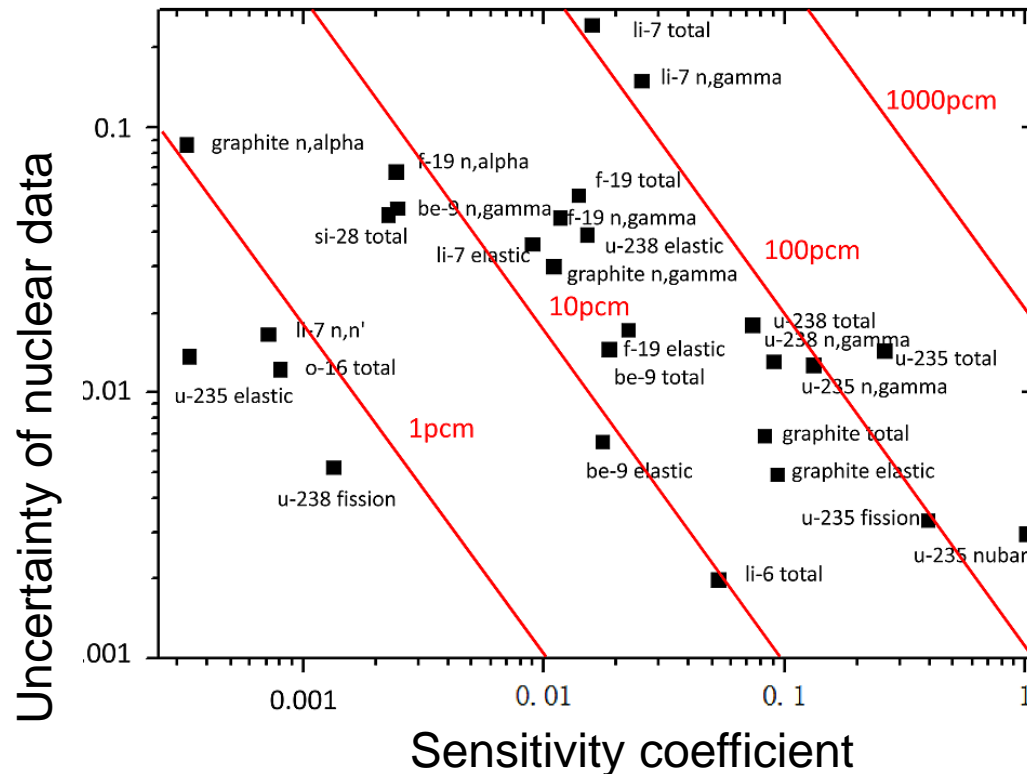
Neutron Spectra

-  Thermal neutron flux raised near reflector.
-  Fast neutron flux dropped very fast near reflector.
-  Because control rod is inserted, peak of power and flux shifts to the lower active core.

Uncertainty and sensitivity

📖 Uncertainty of k_{eff} caused by uncertainties of nuclear data is about 0.58%

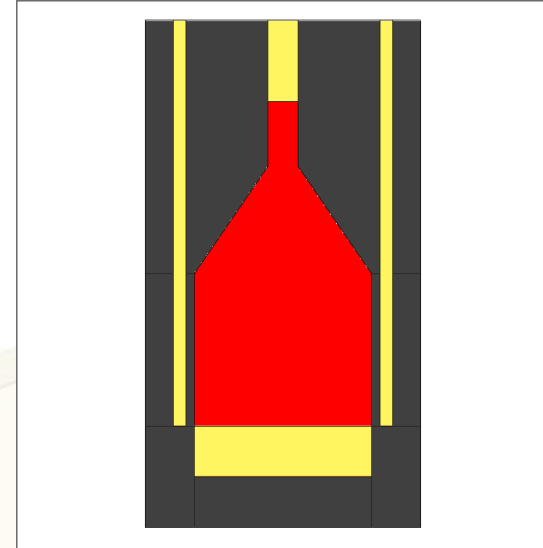
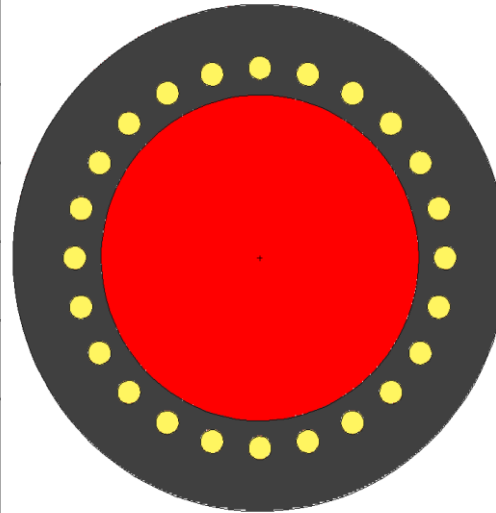
📖 $Li7$ (n, γ) cross-section causes max uncertainty of k_{eff} , about 0.38%.



📖 Uncertainties of nuclear data, uncertainty of k_{eff} , sensitivity coefficient and their relations are demonstrated in the left figure.

Pre-conceptual design of 100MW demonstrated reactor

Parameters	Value
Thermal Power	100 MW
Electric Power	45 MW
Power density	20 MW/m ³
Pebble diameter	3 cm
Coolant of primary loop	2LiF-BeF ₂
Core in/out T	650°C /700°C
Mass of heavy metal	470 kg
U235 enrichment	15%
Volume of active core	4.9 m ³
Height of active core	1.9 m
Diameter of active core	1.8 m

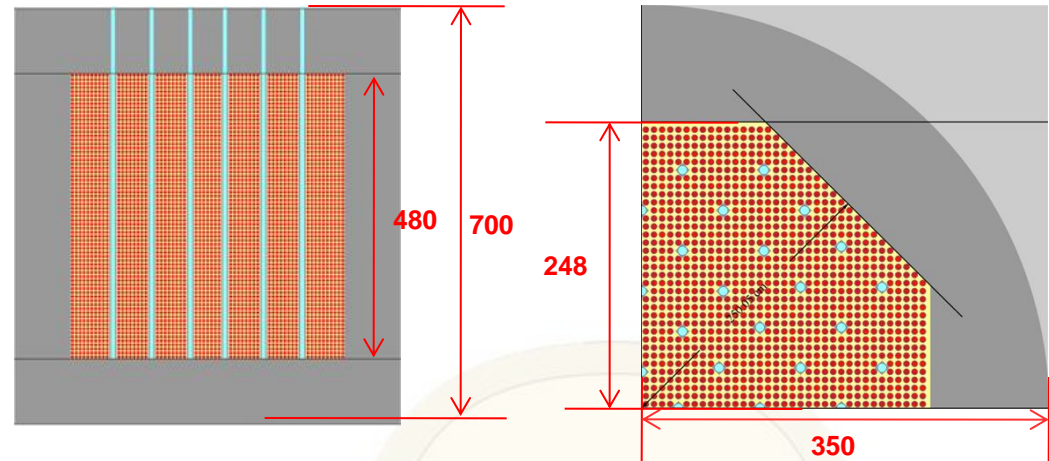


Consideration

- Economic demonstration
- Safety demonstration
- Technology R&D
- Ability of design, building, operation and management

Pre-conceptual design of 1GW order-bed reactor

Parameters	Value
Thermal Power	1 GW
Electric Power	45 MW
Power density	10.1 MW/m ³
Pebble diameter	6 cm
Coolant of primary loop	2LiF-BeF ₂
Core in/out T	600°C / 710°C
Mass of heavy metal	7 t
U235 enrichment	13.5%
Volume of active core	99 m ³
Height of active core	4.8 m
Diameter of active core	~5 m
Max burnup	98 GWD/tU

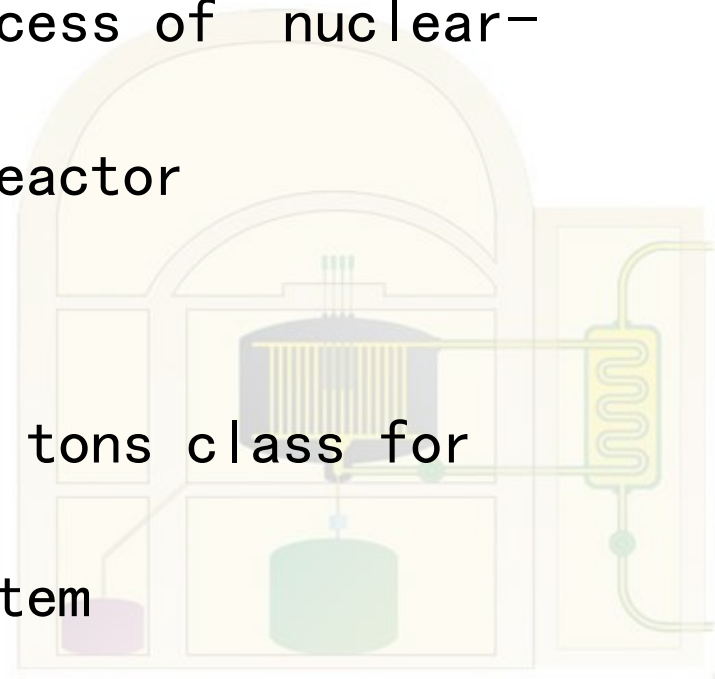


Consideration

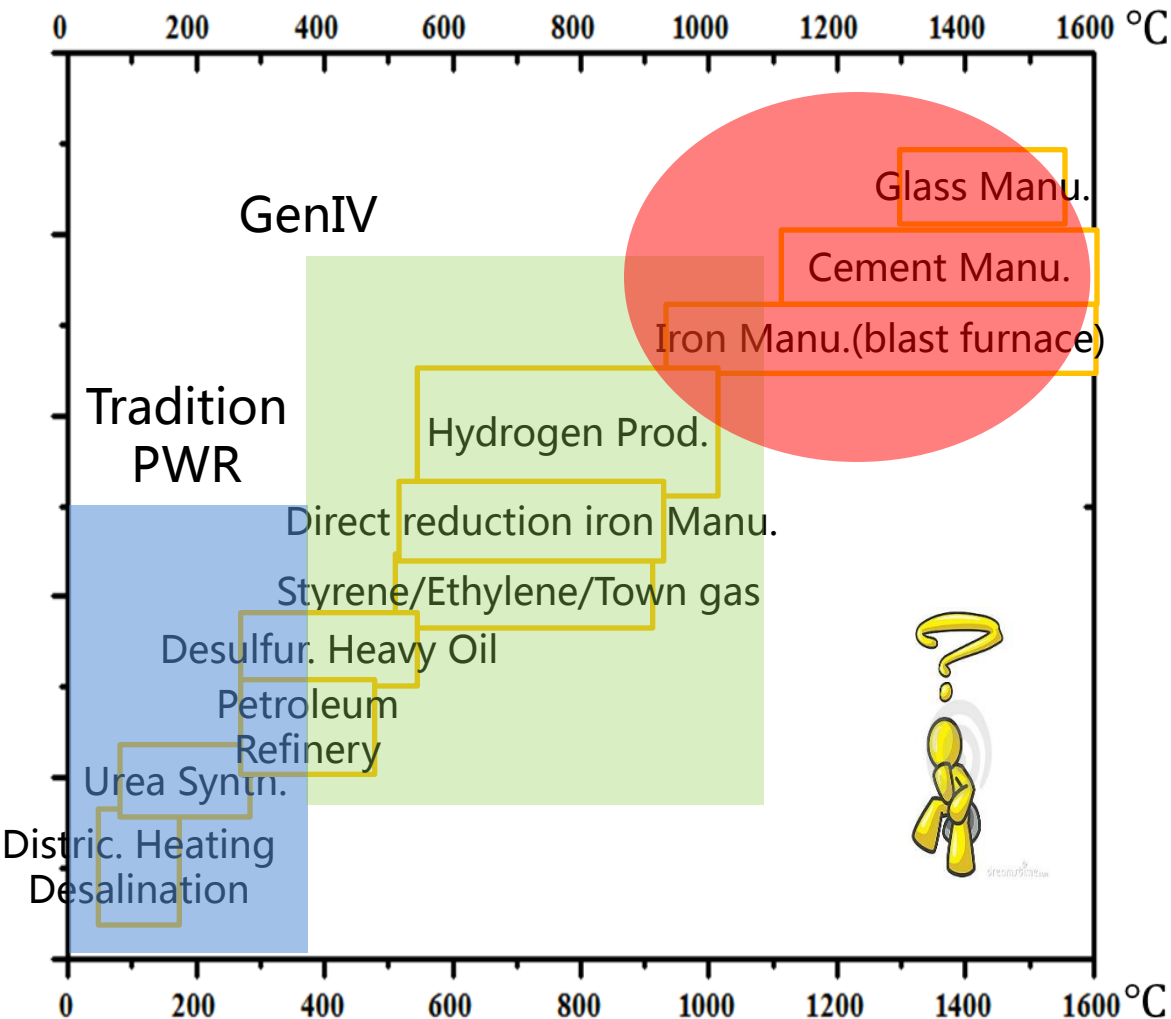
- Ordered-bed
- Utilize current fuel pebble
- Use via hole formed in ordered-bed for reactivity control system

III、Development Plan of Th-based fuel

- Near-term
 - Establish standard specification of nuclear-grade ThF_4 and ThO_2 applicable to MSR
 - Establish the preparation process of nuclear-grade ThF_4 and ThO_2
 - Test Th fuel in molten salt reactor
- Mid-term
 - establish a process line with tons class for nuclear-grade ThF_4 and ThO_2
 - Establish Th-U fuel cycle system



Challenges facing by Nuclear fuel



Challenge

Demand

**Higher
Temp**

**Stable at
HT**

**Deeper
Burnup**

**Radiation
Resistance**

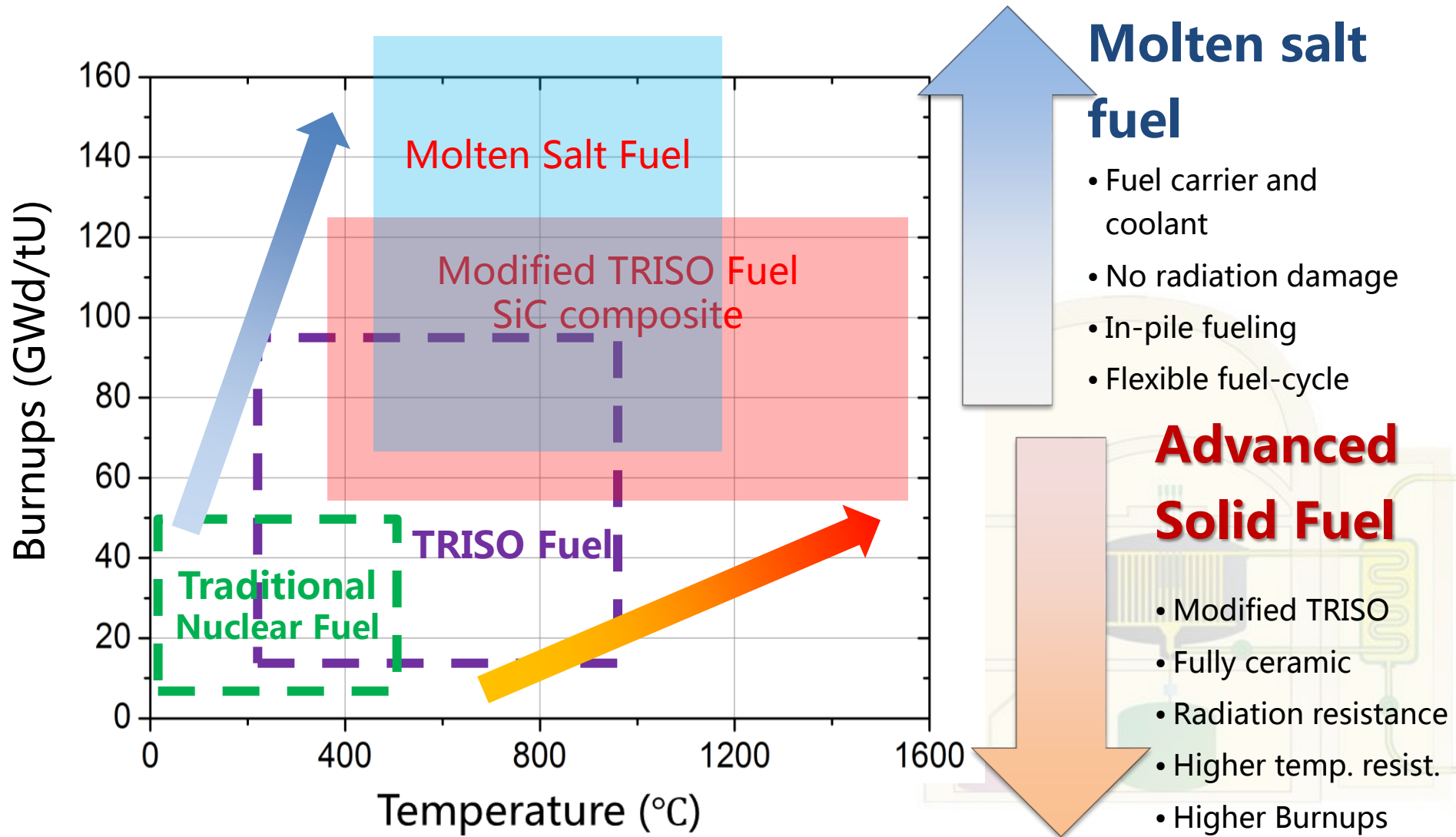
**Higher
Power
Density**

Better Eco.

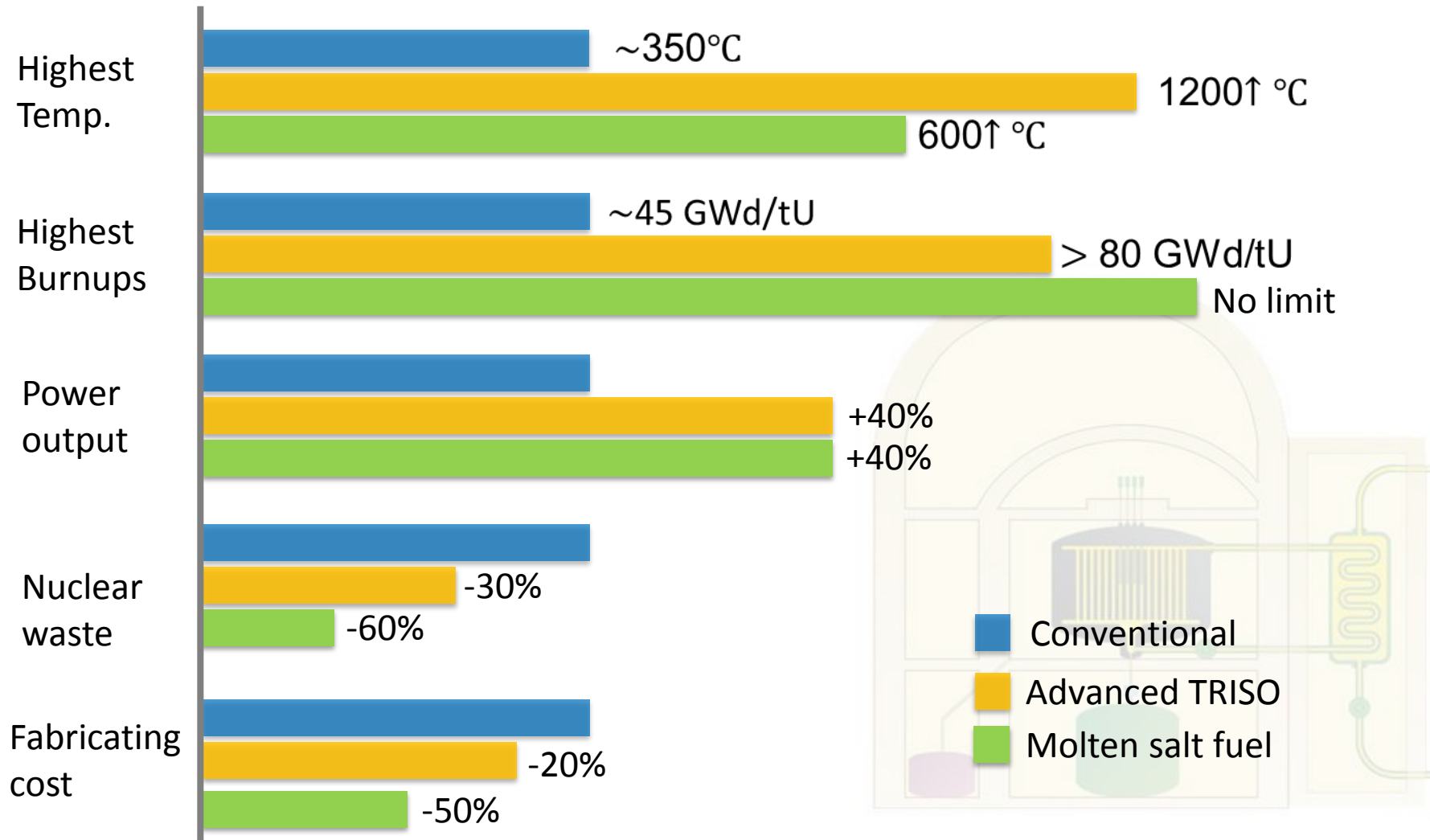
**Limited
Uranium**

**Fuel cycle
Th utilization**

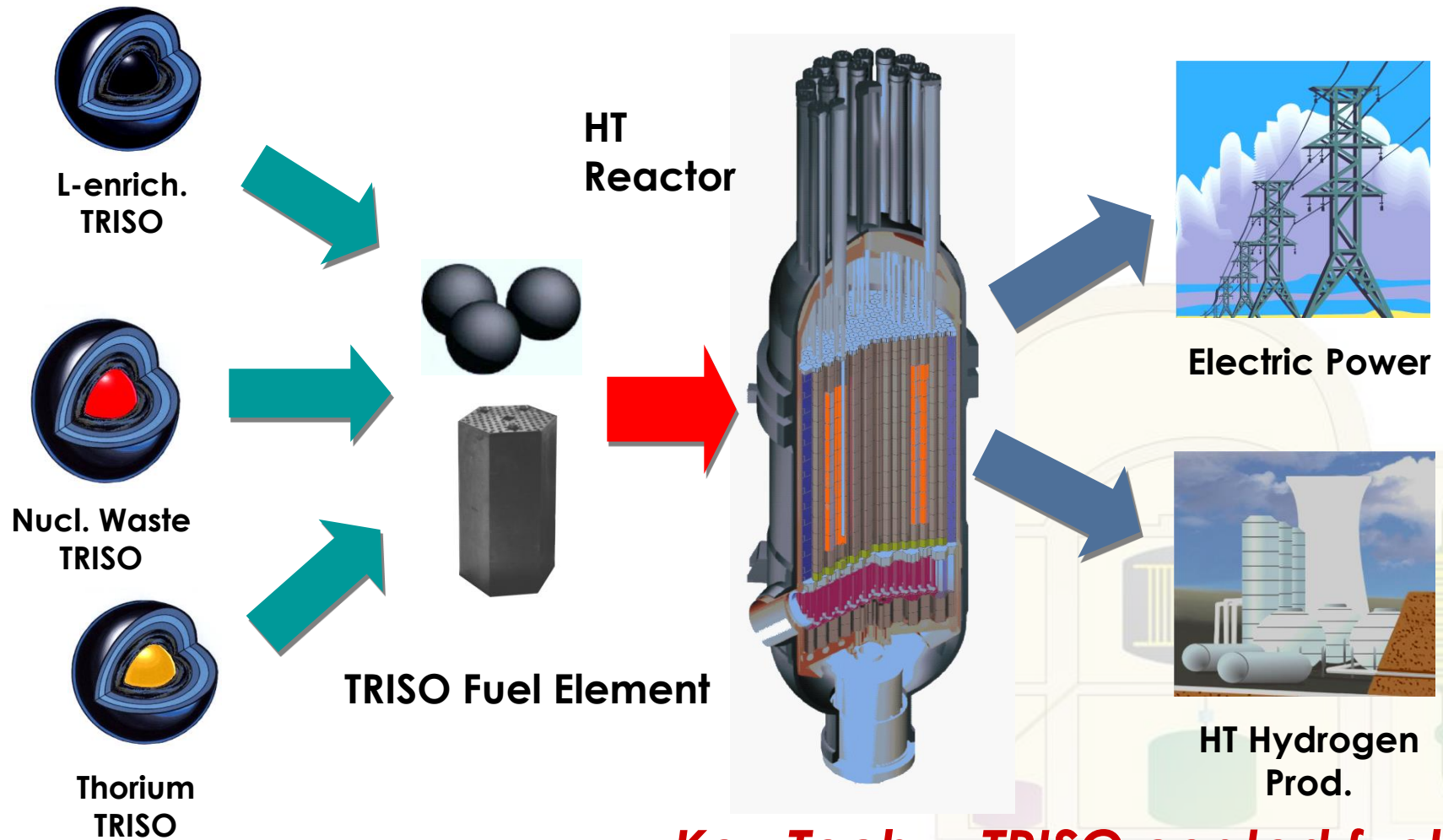
Advanced Fuel Development Map



Aims of Advanced Fuel

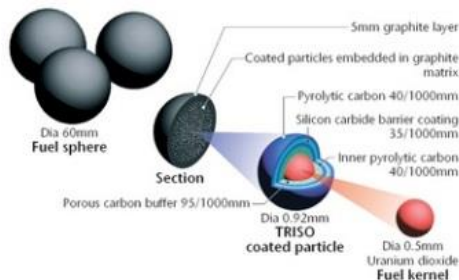
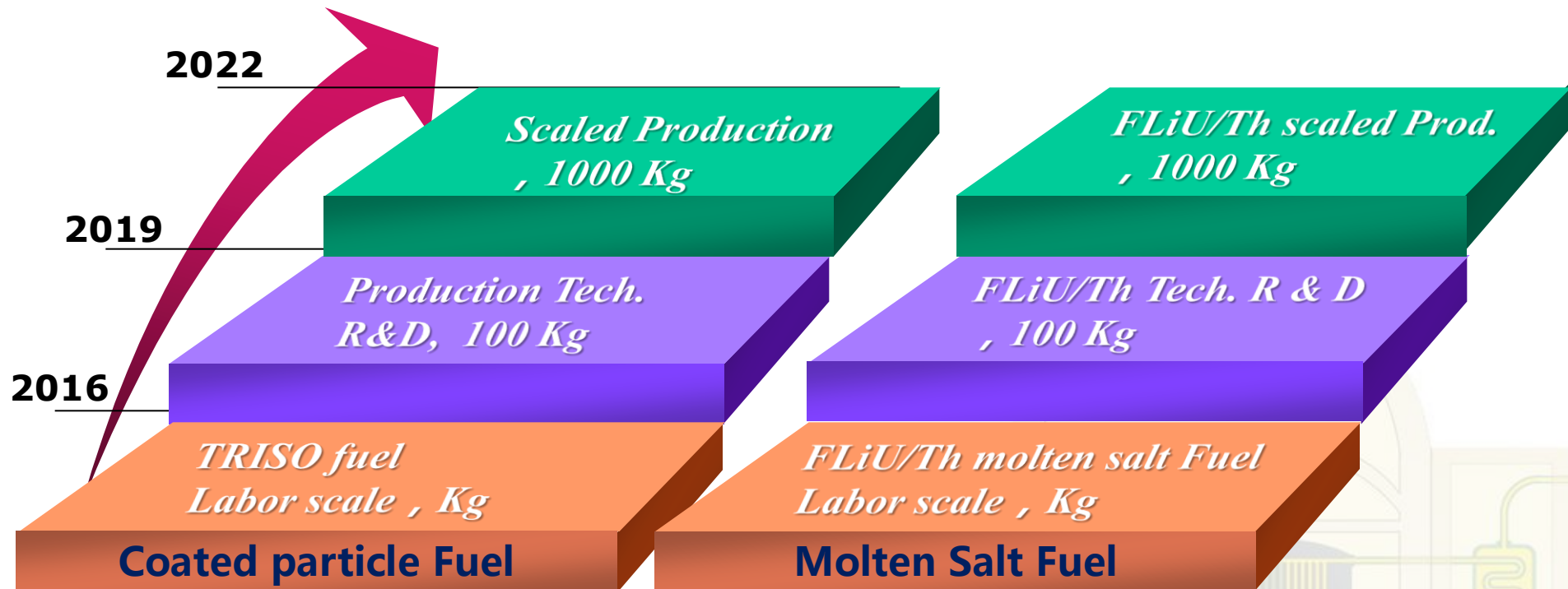


TRISO Fuel: Stand High Temp. and Flexible

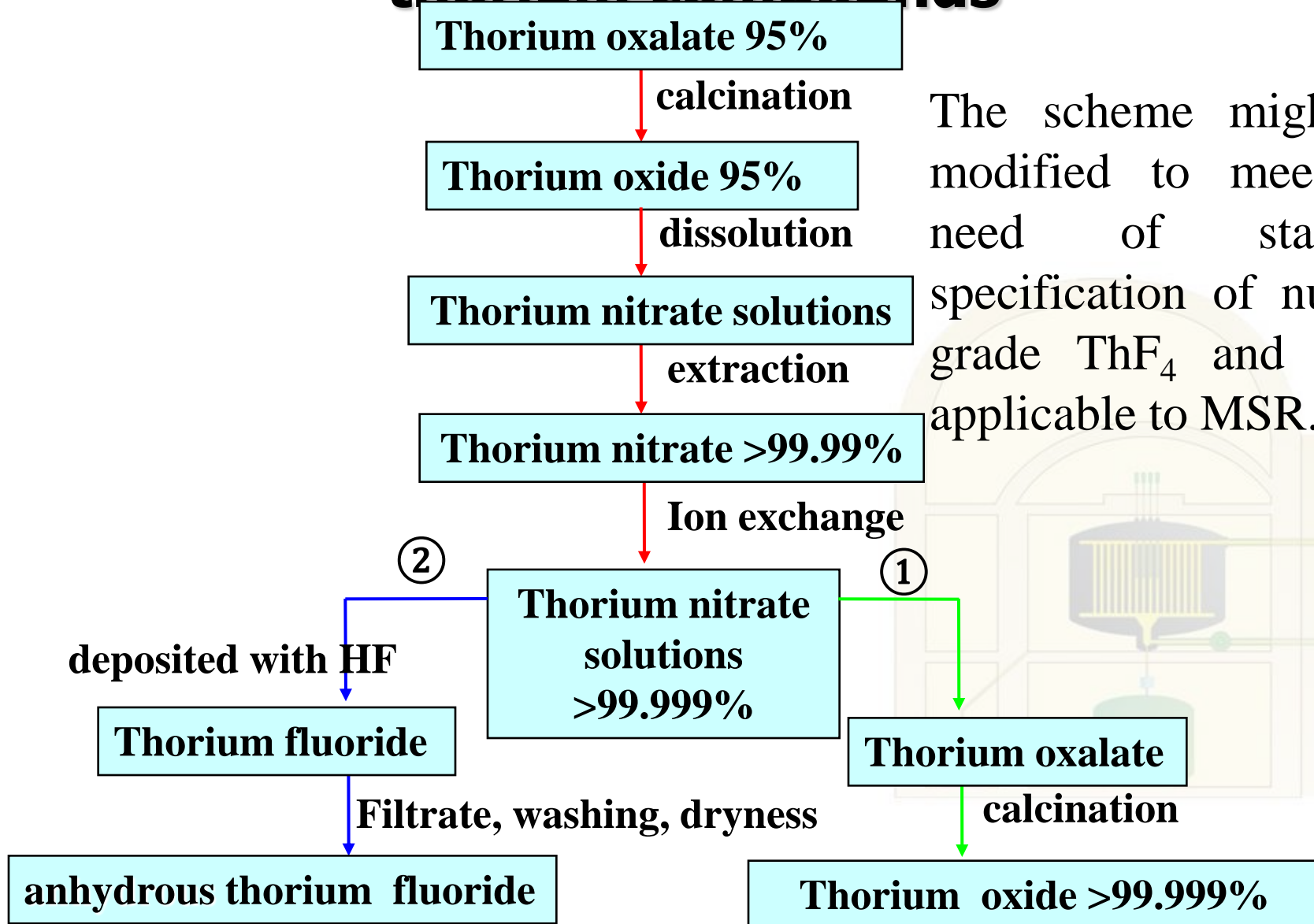


Key Tech: TRISO coated fuel

Development planning for Advanced Fuel



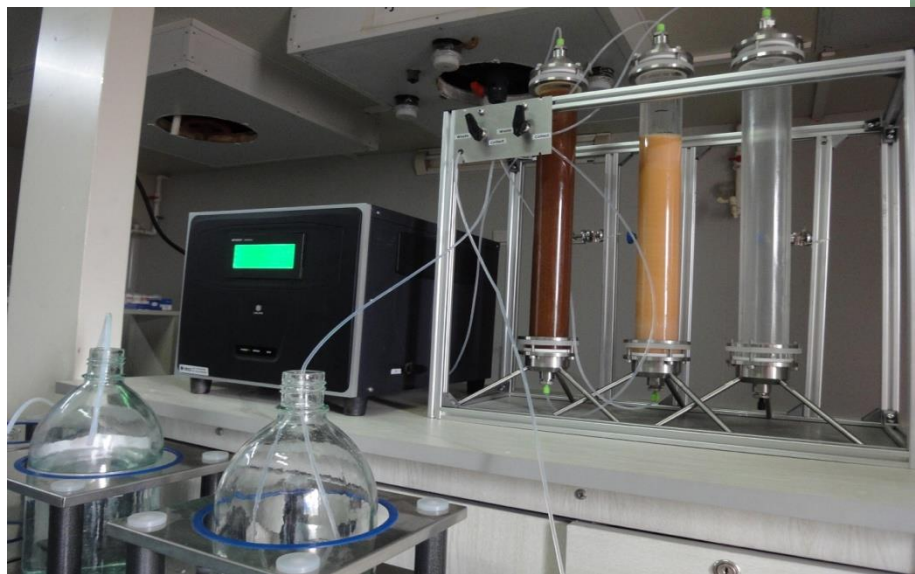
A scheme for the preparation of ultra-pure thorium compounds



The scheme might be modified to meet the need of standard specification of nuclear grade ThF_4 and ThO_2 applicable to MSR.

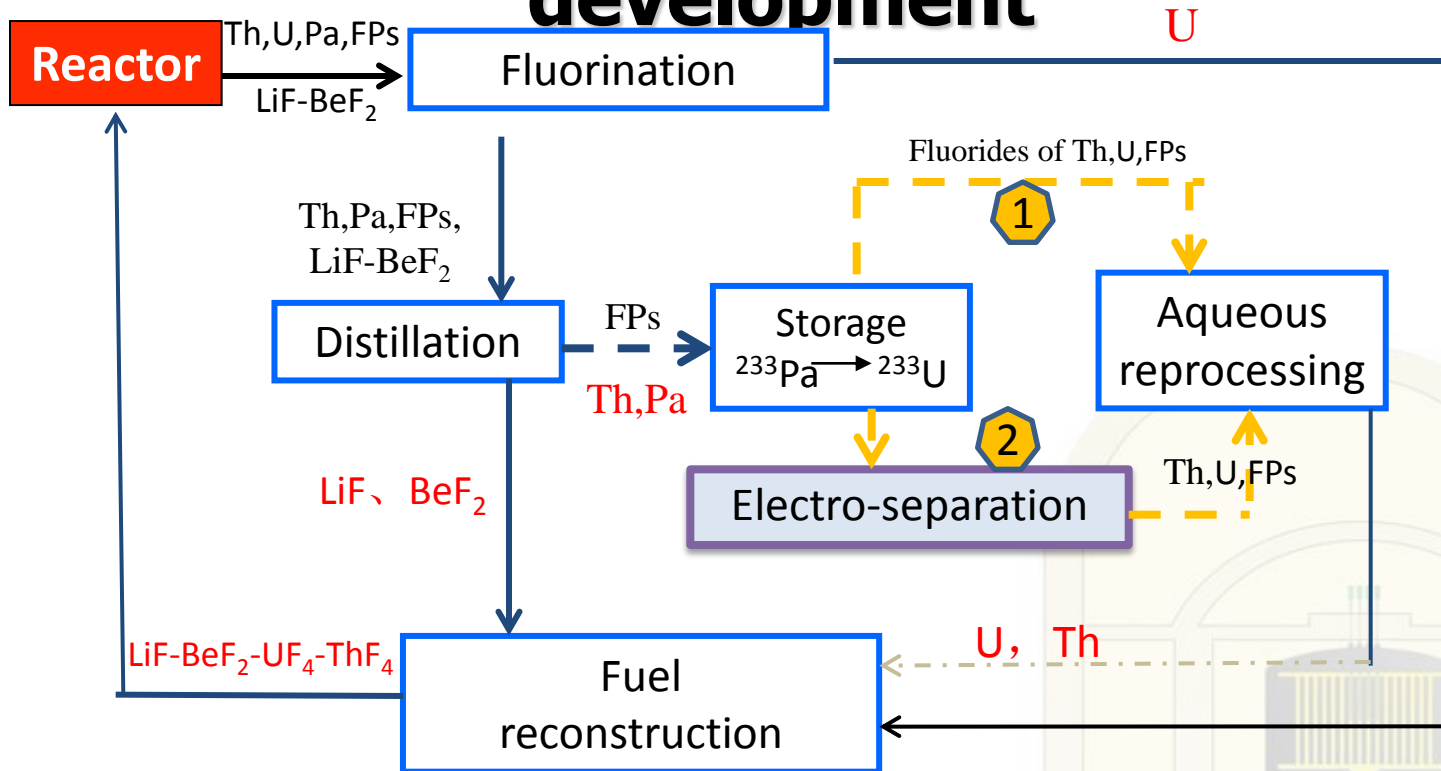
Key techniques developed for ultra-pure thorium compounds preparation

Cascade centrifugal extractors
(50-100g/h)
(99.99%)



Ion exchange column separation system
(200g/h)
(99.999%)

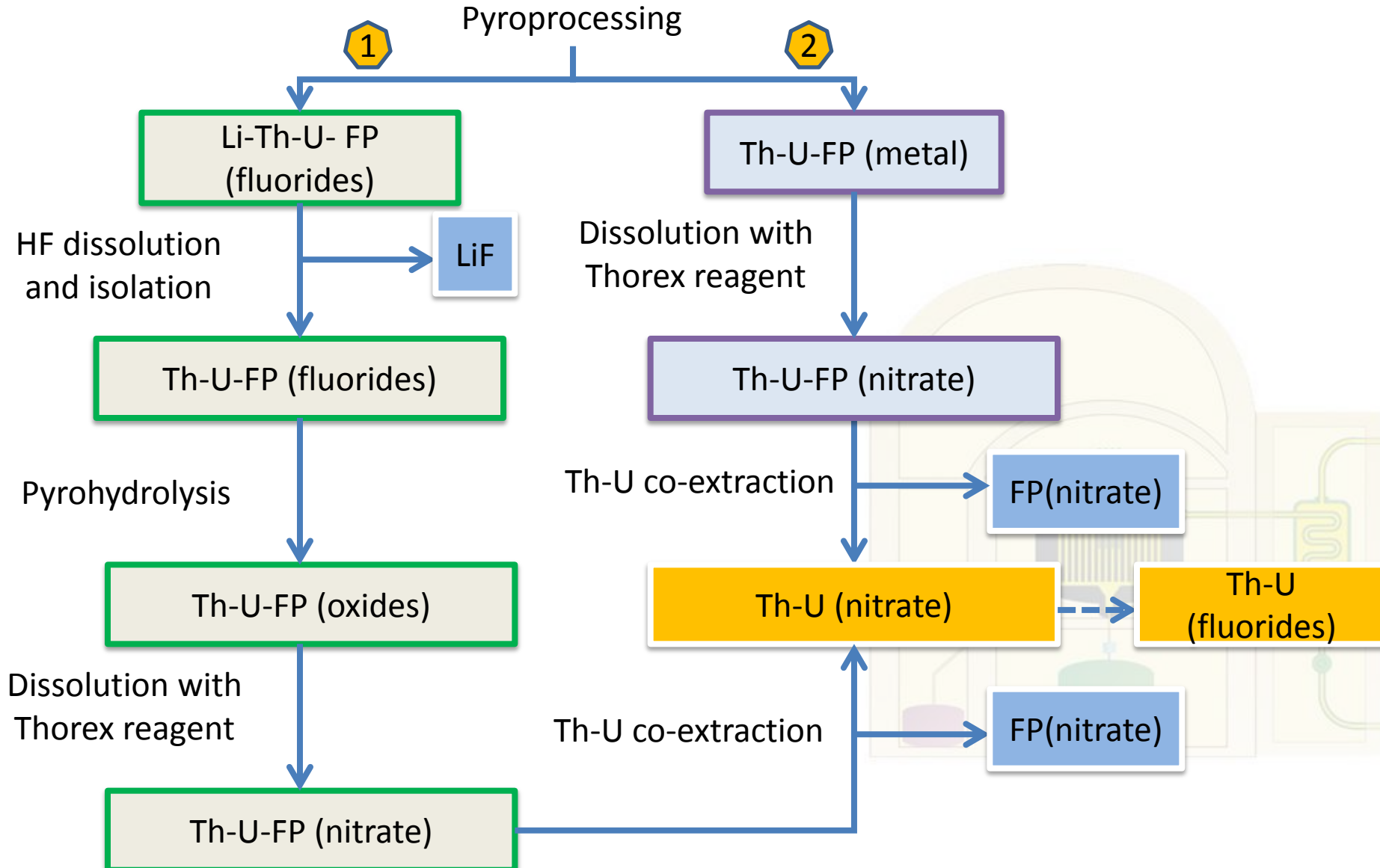
IV-Flow-sheet of TMSR fuel cycle under development



Main advantages:

- ❑ Recycle U and carrier salt as soon as possible based on pyroprocessing to minimize the out-of-reactor inventory and maximize the utilization of fuel
- ❑ Simplify the process by combining with aqueous processing methods
- ❑ **Proliferation risks are negated by co-extraction of Th and U**

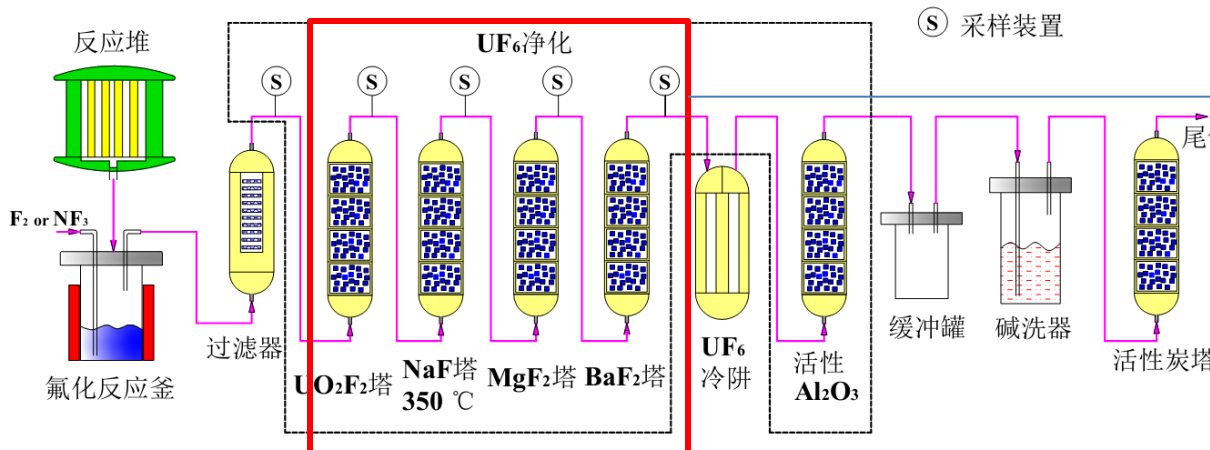
Aqueous Flow-Sheet Following Pyroprocessing



Feasibility study of fluoride volatility process

Key design features

- ▣ Main material: HC-276
- ▣ Volume: 1L
- ▣ Max. Temperature: 600°C
- ▣ F₂ flow: 10ml~1l/min
- ▣ Main constituents:
 - ▣ Gas supply system
 - ▣ Gas preheater
 - ▣ Fluorinator
 - ▣ absorb traps
 - ▣ Off-gas system



Purification of UF₆

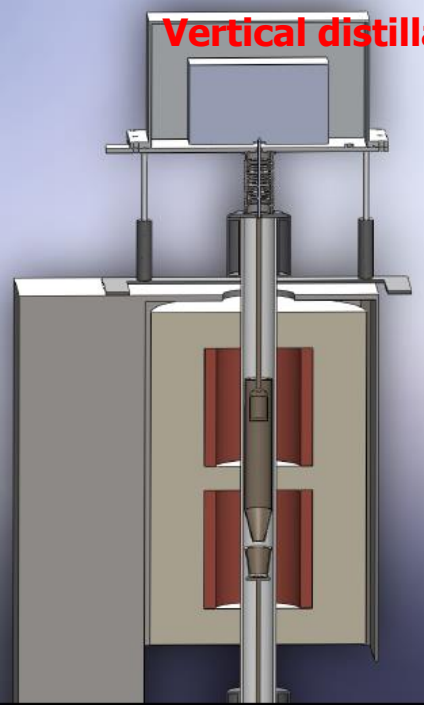
- ▣ UO₂F₂ trap (150 °C) for Pu removal
- ▣ NaF trap (350°C) for Ru and Nb
- ▣ MgF₂ trap (125°C) for Nb, Mo and Sb
- ▣ BaF₂ trap (125°C) for Ru and Mo

Feasibility study of distillation process

◆ Primary results using a simple distillation device showed that carrier salt and rare earth fluorides could be separated efficiently

◆ In present conditions, the highest recovery ratio was about 95% and no significant corrosion was observed

Vertical distillation device



	NdF ₃ residue (%)	Relative volatility of NdF ₃ to LiF*
FLiNaK-NdF ₃ 0.2wt%	83	2.7×10^{-2}
FLiNaK-NdF ₃ 1.6wt%	74	1.3×10^{-2}
FLiNaK-NdF ₃ 3.0wt%	45	2.0×10^{-2}

Residue



Recovery



Dis. vessel



Feasibility study of electrochemical separation process

- ◆ Obtaining a series of electrochemical data for REs, and determination the primarily feasibility of the electrolytic separation for Th and Ln in molten salt

	Reaction	E^0 (V) vs. Ni/NiF ₂	D/cm ² .s ⁻¹	Separation?
Gd	Gd³⁺+3e⁻ → Gd⁰	-2.02	3.2×10^{-4}	Yes
Y	Y³⁺+3e⁻ → Y⁰	-1.96	5.4×10^{-6}	Yes
Nd	Nd³⁺+3e⁻ → Nd⁰	-1.95	1.1×10^{-5}	Yes
Th	Th⁴⁺ +4e⁻ → Th⁰	-1.90	8.4×10^{-7}	Yes
Zr	Zr⁴⁺ +4e⁻ → Zr⁰	-1.86	3.1×10^{-6}	Yes
Zr	Zr⁴⁺ +3e⁻ → Zr³⁺	-1.66	----	---
Sm	Sm³⁺+e⁻ → Sm²⁺	-1.65	7.4×10^{-6}	No
Eu	Eu³⁺+e⁻ → Eu²⁺	-1.03	9.6×10^{-6}	No

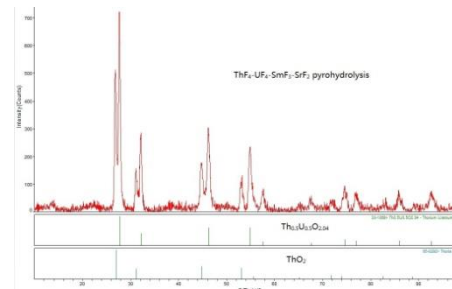
Feasibility study of aqueous process

^7Li recycling by HF Dissolution

Fluorides	Solubility in 90% HF (mg/g)
LiF	90
ThF ₄	<0.03
UF ₄	0.005~0.010

TRE decontamination >500

Pyrohydrolysis



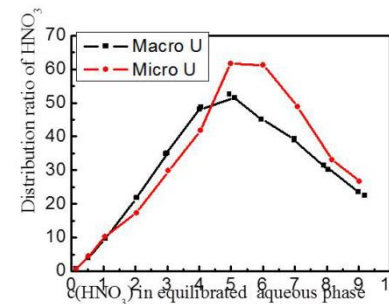
XRD analysis of pyrohydrolysis products of ThF₄-UF₄-SmF₃-SrF₂

ThF₄, UF₄ can be completely converted to oxides @450 °C while RE and Sr, Cs can not.

ThO₂ and Th Metal Dissolution with Thorex reagent

Dissolution of ThO₂ and Th metal with 12-13 M HNO₃ with 0.03-0.05 M fluoride as catalyst is a well known method and has been successfully performed in our lab.

Th-U co-extraction



>99.9% of Th and U can be extracted together with 30% TBP from their solution in 3 M HNO₃ with a decontamination factor of 1E4.



Thank you
for your
Attention!

